

Chemical Formula

Talking about proportion molecules and relative of atoms present (e.g. hydrogen)

Molecular $\text{C}_{10}\text{H}_{22}$ (molecular)

Formula represents the number of atoms of each
in a single molecule

Empirical C_5H_{11} (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit

Atomic Mass

The average mass of atoms present in a sample
unit

Mass

$m = \text{mass}$ unit mass

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Mass

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Note Taking Guide Episode 601 Chemical Formula Answers

Mark Steyvers



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