

Blue cards participate _____ of the possible _____ of atoms _____
 1. _____
 2. _____

Atomic fraction number _____ of particles _____ in the particles _____ of an atom _____

Relative mass an set particles _____ \rightarrow an set particles _____ \rightarrow \rightarrow

Isotopes atoms _____ of an element _____ that have different _____ numbers _____ or _____

Isotopes of Hydrogen

Hydrogen-1

1. _____ protons and _____ neutrons

Hydrogen-2

1. _____ protons and _____ neutrons

Hydrogen-3

1. _____ protons and _____ neutrons

Mass fraction mass _____ number of particles _____ and _____ in an atom _____

Example: Carbon-12 _____ 12.011 _____ 12.011 _____ 12.011 _____

Particle	Charge	Mass	Location
Proton	+1	1.6726 x 10 ⁻²⁷ kg	Nucleus
Neutron	Neutral	1.6749 x 10 ⁻²⁷ kg	Nucleus
Electron	-1	9.109 x 10 ⁻³¹ kg	Electron cloud

Atomic mass average _____ of the masses _____ of all the isotopes of _____

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Note Taking Guide Episode 302 Answers Chemistry

Karin Nielsen-Saines



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