Note Taking Guide: Episode 403

Name _____

Atomic Number

- _Increase_ across a period
- increase down a family

Atomic Mass

- _increase_ across a period
- _increase_ down a family

Atomic Rodius

half the distance between _nuclei_ of two like _atoms_.

- _decreases_ across a period (due to increased attraction of more positive nuclei to electron clouds)
- increase down a family (due to addition of an energy level)

Electron Affinity

amount of _energy_ when an_atom_ gains an

electron

- increases across a period
- _decreases_ down a family

Ionization Energy

amount of _energy_ required to _remove_ an _electron_ from a neutral _atom_

- _increase_ across a period
- _decrease_ down a family

Electron Distribution - Forming Ions

- Atoms become ions by _gaining_ or _losing_ electrons to become _an ion_.
- An atom is stable when it has a _____full_ outer energy level (_8_ valence electrons).
- Atoms that lose (give away) electrons become _positive_ charged ions.
 Atoms that gain (steal) electrons become _negative_ charged ions.

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G Thomas

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