

Quantum Theory

- a. Energy is quantized in $\frac{h\nu}{2\pi}$ and $\frac{h\nu}{2\pi}$ is the minimum energy
- b. Photo electron of energy is emitted if $\frac{h\nu}{2\pi} > \phi$
- c. In quantized light energy is emitted $\frac{h\nu}{2\pi}$
- d. The energy of a photon of light depends on the $\frac{h\nu}{2\pi}$ of light

$$E = \frac{h\nu}{2\pi} \text{ constant}$$

_____ now the question is how to indicate this phenomenon

- a. Light energy of $\frac{h\nu}{2\pi}$ is light having a certain amount of energy depending on the $\frac{h\nu}{2\pi}$ $h\nu = E$
- b. Since $\frac{h\nu}{2\pi}$ is constant a single photon of light
- c. It also has certain amount of $\frac{h\nu}{2\pi}$ so group can do this work
- d. If $\frac{h\nu}{2\pi}$ and $\frac{h\nu}{2\pi}$ energy groups, this electron will be ejected
- e. If it is constant then energy group that electron needs to escape will be ϕ and the electron will be $\frac{h\nu}{2\pi}$ and $\frac{h\nu}{2\pi}$ energy group

What happens for light of constant light?

(a)

(b)

_____ $\frac{h\nu}{2\pi}$

_____ $\frac{h\nu}{2\pi}$ and $\frac{h\nu}{2\pi}$

Note Taking Guide 1302 Answers

G Thomas



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