ALKENES:

beneral formula = Cn.Hzm.

functional group: c=C(saturated)

REACTIONS

Hydration (420) => +-c-c-+

- outsides contrast - migh pressurer temp. - Forms excond

Halogenarion(Brz) => H

Hydrogenation(42) =7

- michiga coaronyst - Tempilso" - forms a licane

compustion = 7 complete: (02th20 incomplete: (0th20

WITCOINO IT:

functional group OH

General formula = Callenn OH

REACTIONS

Product: (0; +420 Burning =>

heaverouses. Add water (470) => (no H+ or OH-)

Add oxygen (02) = 7 forms (armoxytic audidi

, sugar t) yeast 3) mildion formentation => Ethanoll +) Time . . filtered remove yeast disn't remove water.

CABOXYLIC ACIDS:

functional group = COOH

General formula: (mHznerCOOH.

REACTIONS

Acid+Water=> (H3(00H= (H3(00+H+ weak acid-partially lonised w

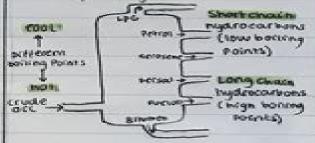
(Sandillaren (O.Decinos IV)) Acid+ (asbonate => (B3(00H+ Nas(0) -> (Dg+H10+CH3(00Nh (aq) due to CO)

ACCO + ALCOHOL -> CH3COOH + CH3CH2OH -> CH3CCOCH3CH2 + TH2O

eq emanoic acid+emanol > emyl emanoate.

FRACTIONAL DOSTULATION:

used to apperous mustocarbon fractions.



POLYMERISATION:

one monomer Addition Polymentsation

Condensation polymerisation . water side product

)-c-0H + HP

PO MOCO . 40- C-O-C-OH + H-N-D-N-H (N-N-D-NH)

- 12-0-8-M-10-M-1 +MEO

CRACKING

The breaking up the hydrocarbon chains which are too long from fractional distribution into shorter more oterous hydrocarbons.

TWO TYPES

O steam cracking: Vaporised hydrocarbon't steam 3-08 (Cataly) recovering: varioused by a sociation + catalyst.

Example:

Calle HGHIN Calling =7

One alkene one alkane.

(7 H14+(H4)

6 HIZT CZHA

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BM King

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