

### Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular  $C_2H_6O$  (molecules)

Formula represents the number of \_\_\_\_\_ of each  
in a single molecule.

Empirical  $C_2H_6O$  (molecules)

Formula represents the simplest ratio of  
positive and negative ions in a formula  
unit.

### Atomic Mass

The average mass of an atom, weighted  
average.

### Isotopes

$^{12}C$  and  $^{13}C$  isotopes

Isotopes are chemically identical  
atoms.

### Abundance

$^{12}C$  abundance

Isotopes are chemically identical

$^{12}C$

# Note Taking Guide Physics 601

**Siddappa N.Byrareddy**



## **Note Taking Guide Physics 601:**

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