

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equated _____

2. $40g = 40r + 4 \cdot 20g$

2. Magnesium nitride reacts with a single oxygen when the products of synthesis are at magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ reaction in the _____

3. Problem

When aluminum aluminum reacts with aluminum oxide, aluminum oxide is produced



moles of aluminum

$\frac{1}{2}$	Al	$\frac{3}{2}$	O_2
$\frac{1}{2}$	Al	$\frac{3}{2}$	Al_2O_3
$\frac{1}{2}$	O_2	$\frac{3}{2}$	Al_2O_3

2. If you start with 4 moles of Al , how many moles of Al_2O_3 will be produced?

2 moles Al (moles Al_2O_3) and $3g = 4$ and $3g$

Note Taking Guide Episode 801 Stoichiometry Answers

CH Cherryholmes



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Note Taking Guide Episode 801 Stoichiometry Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Note Taking Guide Episode 801 Stoichiometry Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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