

Worksheet #1: Significant Figures & Dimensional Analysis

Directions: For each problem below, write the equation and show your work. Be sure to **box your final answer**.

Part 1: Solve the following problems using scientific notation and rounding to the appropriate value.

- a. $300,000,000 \rightarrow 3.0 \times 10^8$
- b. $450,000 \rightarrow 4.5 \times 10^5$
- c. $0.000000000 \rightarrow 9.0 \times 10^{-9}$
- d. $-0.0000000 \rightarrow -3.0 \times 10^{-6}$

Part 2: Solve the following problems, and show your final answer with the appropriate number of significant figures.

- a. $1.2000 + 71.80 = 73.0000 \rightarrow 73.00$
- b. $9.1 \times 10^3 + 2.1 \times 10^3 = 11.2 \times 10^3 = 1.12 \times 10^4$
- c. $0.000000000 + 0.000000000 = 0.000000000$
- d. $0.000000000 + 0.000000000 = 0.000000000$
- e. $0.000000000 + 0.000000000 = 0.000000000$
- f. $0.000000000 + 0.000000000 = 0.000000000$
- g. $0.000000000 + 0.000000000 = 0.000000000$
- h. $0.000000000 + 0.000000000 = 0.000000000$
- i. $0.000000000 + 0.000000000 = 0.000000000$
- j. $0.000000000 + 0.000000000 = 0.000000000$
- k. $0.000000000 + 0.000000000 = 0.000000000$
- l. $0.000000000 + 0.000000000 = 0.000000000$
- m. $0.000000000 + 0.000000000 = 0.000000000$
- n. $0.000000000 + 0.000000000 = 0.000000000$
- o. $0.000000000 + 0.000000000 = 0.000000000$
- p. $0.000000000 + 0.000000000 = 0.000000000$
- q. $0.000000000 + 0.000000000 = 0.000000000$
- r. $0.000000000 + 0.000000000 = 0.000000000$
- s. $0.000000000 + 0.000000000 = 0.000000000$
- t. $0.000000000 + 0.000000000 = 0.000000000$
- u. $0.000000000 + 0.000000000 = 0.000000000$
- v. $0.000000000 + 0.000000000 = 0.000000000$
- w. $0.000000000 + 0.000000000 = 0.000000000$
- x. $0.000000000 + 0.000000000 = 0.000000000$
- y. $0.000000000 + 0.000000000 = 0.000000000$
- z. $0.000000000 + 0.000000000 = 0.000000000$

Part 3: For each problem below, show your work. Always use units and box in your final answer.

- a. The density of pure silver is 10.5 g/cm^3 at 20°C . If 5.25 g of pure silver pellets are added to a graduated cylinder containing 11.2 mL of water, to what volume level will the water in the cylinder rise?

$d = \frac{m}{V}$

$d = 10.5 \text{ g/cm}^3$ $m = 5.25 \text{ g}$ $V = ?$

$V = \frac{m}{d} = \frac{5.25 \text{ g}}{10.5 \text{ g/cm}^3} = 0.5 \text{ cm}^3$

$11.2 \text{ mL} + 0.5 \text{ mL} = 11.7 \text{ mL}$

- b. The density of air at ordinary atmospheric pressure and 20°C is 1.18 g/L . What is the mass, in kilograms, of the air in a room that measures $12.5 \text{ m} \times 15.5 \text{ m} \times 3.0 \text{ m}$?

$d = 1.18 \text{ g/L}$ $V = 12.5 \times 15.5 \times 3.0 = 581.25 \text{ m}^3$ $m = ? \text{ kilograms}$

$1 \text{ m}^3 = 1000 \text{ L}$

$581.25 \text{ m}^3 \times 1000 = 581,250 \text{ L}$

$m = d \times V = 1.18 \text{ g/L} \times 581,250 \text{ L} = 685,875 \text{ g}$

$685,875 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 685.875 \text{ kg}$

Final Answer: 685.875 kg

$20.317 \text{ g} \times 1000 = 20,317 \text{ g}$

$20,317 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 20.317 \text{ kg}$

Northridge Learning Center Packet Answers Chemistry

D Keegan



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web oct 21 2019 utilisation du bleu de prusse dans la peinture malgré sa teinte bleu unique le bleu de prusse n est pas une couleur qui possède que des qualités cette teinte est réputée pour son manque de solidité et il est par ailleurs très difficile de la mélanger avec d autres couleurs car elle à tendance à les salir lorsqu elle entre en contact avec

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web la première couleur artificielle le bleu de prusse page précédente au début du xviii e siècle la seule teinte bleue stable est le bleu outremer obtenu à partir d une pierre précieuse le lapis lazuli cette couleur bleu est donc très chère

fiche piratome n 2 bleu de prusse - Mar 07 2022

web le bleu de prusse insoluble dans l eau est un agent chélateur qui possède une très forte affinité pour le césium et le thallium en agissant par échanges d ions les complexes bleu de prusse césium ou bleu de prusse thallium ainsi formés dans le tube digestif lors des intoxications aiguës sont peu