

Sample paper - 1

GENERAL INSTRUCTION

1. You will not be allowed to write during the 15 minutes. This time is being spent in reading the question paper.
2. The time given at the head of this paper is the time allowed for writing the answer.
3. Attempt all questions from Section A and any 4 questions from Section B.
4. The intended marks for questions or parts of questions are given in brackets []

Time: 2Hrs
Marks:80

Max.

SECTION A [40 marks]

1.
 - (a) Fill in the blanks from the choices given in brackets.
 - (i) An element with atomic number 17 belongs to.....period.(first, third)
Sol:
Third
 - (ii) Write the electronic configuration of magnesium atom after losing 2 electrons.....(neon, argon)
Sol:
Neon
 - (iii) Whose example is ammonium hydroxide.....? (weak, strong)
Sol:
Weak
 - (iv) Write the oxidized product when the H_2S reacts with dil. HNO_3
(sulphur, nitrogen)
Sol:
Sulphur
 - (v) Write the molecular formula of the compound when ethyne reacts with chlorine ($C_2H_4Cl_2$, $C_2H_2Cl_2$)
Sol:
 $C_2H_2Cl_2$
 - (b) Choose the correct answer from the option
 - (i) Which law groups the elements in family of threem
(a) Law of traids

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Table of Contents Sample Papers Icse Board Class 1

1. Understanding the eBook Sample Papers Icse Board Class 1
 - The Rise of Digital Reading Sample Papers Icse Board Class 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Papers Icse Board Class 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Papers Icse Board Class 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Papers Icse Board Class 1
 - Personalized Recommendations
 - Sample Papers Icse Board Class 1 User Reviews and Ratings
 - Sample Papers Icse Board Class 1 and Bestseller Lists
5. Accessing Sample Papers Icse Board Class 1 Free and Paid eBooks

- Sample Papers Icse Board Class 1 Public Domain eBooks
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 - Sample Papers Icse Board Class 1 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Papers Icse Board Class 1
 - Highlighting and Note-Taking Sample Papers Icse Board Class 1
 - Interactive Elements Sample Papers Icse Board Class 1
- 8. Staying Engaged with Sample Papers Icse Board Class 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Papers Icse Board Class 1
- 9. Balancing eBooks and Physical Books Sample Papers Icse Board Class 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Papers Icse Board Class 1
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sample Papers Icse Board Class 1
 - Setting Reading Goals Sample Papers Icse Board Class 1
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sample Papers Icse Board Class 1
 - Fact-Checking eBook Content of Sample Papers Icse Board Class 1
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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