(a) Interpret and use the terms:

(i) Empirical formula = simplest whole number ratio of atoms of each element present in a compound

(ii) Molecular formula = actual number of atoms of each element in a molecule,

(iii) General formula = simplest algebraic formula of a member of a homologous series

Alkane	CoH2n+2
Alkene	C-Ha-
Alcohol	C-H22OH
Cycloalkane	C-H _{2m}
Cycloalicene	Call Page 1
Haloalkane .	C-HX

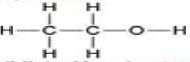
(iv) Structural formula = minimal detail that shows the arrangement of atoms in a molecule

E.g. Butane:

CH3CH2CH2CH3 or CH3(CH2)2CH3,

(v) Displayed formula = relative positioning of atoms and the bonds between them,

i.e. for ethanol:



(vi) Skeletal formula = simplified organic formula, shown by removing hydrogen atoms from alkyl chains, leaving just a carbon skeleton and associated functional group

ie for butan-1-ol:



(b) Interpret, and use, the terms:

(i) hamologous series = a series of organic compounds having the same functional group but with each successive member differing by CH2.

(ii) functional group = a group of atoms responsible for the characteristic reactions of a compound;

(c) Use the general formula of a homologous series to predict the formula of any member of the series;

E.g - Alkane = C_nH_{2n+2} If C = 9, $H = (9 \times 2) + 2$

Therefore = C_9H_{20} = Nonane

(d) State the names of the first ten members of the alkane homologous series;

No of Carbons	Alkyl Group
	Methyl
2	Ethyl
3	Propyl
4	Butyl
5	Pentyl
6	Hiescyl
7	Heptyl
8	Octyl
9	Nonyl
10	Decyl

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Table of Contents Ocr Chemistry Unit F32june 2014 Past Paper

- 1. Understanding the eBook Ocr Chemistry Unit F32june 2014 Past Paper
 - The Rise of Digital Reading Ocr Chemistry Unit F32june 2014 Past Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Ocr Chemistry Unit F32june 2014 Past Paper
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Chemistry Unit F32june 2014 Past Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Ocr Chemistry Unit F32june 2014 Past Paper
 - Personalized Recommendations
 - o Ocr Chemistry Unit F32june 2014 Past Paper User Reviews and Ratings
 - Ocr Chemistry Unit F32june 2014 Past Paper and Bestseller Lists
- 5. Accessing Ocr Chemistry Unit F32june 2014 Past Paper Free and Paid eBooks
 - Ocr Chemistry Unit F32june 2014 Past Paper Public Domain eBooks
 - Ocr Chemistry Unit F32june 2014 Past Paper eBook Subscription Services
 - o Ocr Chemistry Unit F32june 2014 Past Paper Budget-Friendly Options
- 6. Navigating Ocr Chemistry Unit F32june 2014 Past Paper eBook Formats
 - o ePub, PDF, MOBI, and More
 - o Ocr Chemistry Unit F32june 2014 Past Paper Compatibility with Devices
 - Ocr Chemistry Unit F32june 2014 Past Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Chemistry Unit F32june 2014 Past Paper
 - o Highlighting and Note-Taking Ocr Chemistry Unit F32june 2014 Past Paper
 - Interactive Elements Ocr Chemistry Unit F32june 2014 Past Paper
- 8. Staying Engaged with Ocr Chemistry Unit F32june 2014 Past Paper

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Ocr Chemistry Unit F32june 2014 Past Paper
- 9. Balancing eBooks and Physical Books Ocr Chemistry Unit F32june 2014 Past Paper
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Ocr Chemistry Unit F32june 2014 Past Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr Chemistry Unit F32june 2014 Past Paper
 - Setting Reading Goals Ocr Chemistry Unit F32june 2014 Past Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr Chemistry Unit F32june 2014 Past Paper
 - Fact-Checking eBook Content of Ocr Chemistry Unit F32june 2014 Past Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - \circ Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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