# Reference Tables for Physical Setting/EARTH SCIENCE

#### Radioactive Decay Data

| RADIOACTIVE<br>ISOTOPE | DISINTEGRATION                       | HALF-LIFE<br>(years)   |
|------------------------|--------------------------------------|------------------------|
| Carbon-14              | 14C →14N                             | $5.7 \times 10^{3}$    |
| Potassium-40           | <sup>40</sup> K → <sup>40</sup> Ar   | 1.3 × 10 <sup>9</sup>  |
| Uranium-238            | 236 <sub>U</sub> → 206 <sub>Pb</sub> | 4.5 × 10 <sup>9</sup>  |
| Rubidium-87            | <sup>87</sup> Rb <sup>87</sup> Sr    | 4.9 × 10 <sup>10</sup> |

#### Equations

|                       | stance between foci<br>ength of major axis |
|-----------------------|--|
|                       | ge in field value<br>distance              |
| Rate of change =      | change in value<br>time                    |
| Density = mass volume | 77 8 28 P. 27 C. S.                        |

#### Specific Heats of Common Materials

| MATERIAL          | SPECIFIC HEAT<br>(Joules/gram + "C) |  |  |
|-------------------|-------------------------------------|--|--|
| Liquid water      | 4.18                                |  |  |
| Solid water (ice) | 2.11                                |  |  |
| Water vapor       | 2.00                                |  |  |
| Dry air           | 1.01                                |  |  |
| Basalt            | 0.84                                |  |  |
| Granite           | 0.79                                |  |  |
| Iron              | 0.45                                |  |  |
| Copper            | 0.38                                |  |  |
| Lead              | 0.13                                |  |  |

#### Properties of Water

| Heat energy gained during melting 334 J/g         |
|---|
| Heat energy released during freezing 334 J/g      |
| Heat energy gained during vaporization 2260 J/g   |
| Heat energy released during condensation 2260 J/g |
| Density at 3.98°C                                 |

#### Average Chemical Composition of Earth's Crust, Hydrosphere, and Troposphere

| ELEMENT<br>(symbol) | CRUST           |                   | HYDROSPHERE       | TROPOSPHERE       |
|---------------------|-----------------|-------------------|-------------------|-------------------|
|                     | Percent by mass | Percent by volume | Percent by volume | Percent by volume |
| Oxygen (O)          | 46.10           | 94.04             | 33.0              | 21.0              |
| Silicon (Si)        | 28.20           | 0.88              |                   |                   |
| Aluminum (Al)       | 8.23            | 0.48              |                   |                   |
| Iron (Fe)           | 5.63            | 0.49              |                   |                   |
| Calcium (Ca)        | 4.15            | 1.18              |                   |                   |
| Sodium (Na)         | 2.36            | 1.11              |                   |                   |
| Magnesium (Mg)      | 2.33            | 0.33              |                   |                   |
| Potassium (K)       | 2.09            | 1.42              |                   | 5-0-0-0-0-0       |
| Nitrogen (N)        |                 |                   |                   | 78.0              |
| Hydrogen (H)        |                 |                   | 66.0              |                   |
| Other               | 0.91            | 0.07              | 1.0               | 1.0               |

#### 2011 EDITION

This edition of the Earth Science Reference Tables should be used in the classroom beginning in the 2011–12 school year. The first examination for which these tables will be used is the January 2012 Regents Examination in Physical Setting/Earth Science.

Eurypterus remipos

New York State Fossil

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**Liying Dong** 

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