Question			Answer	Mark	Guidance
5	(a)		$\begin{array}{cccccccccccccccccccccccccccccccccccc$	2	Displayed formulae MUST be used to award each mark
			Balanced equation for formation of correct polymer - correct use of n in the equation and brackets ✓		n on LHS can be at any height to the left of formula  AND n on the RHS must be a subscript (essentially below the side link)
Ī	(b)	(1)	CH <sub>2</sub> CHC/ +2O <sub>2</sub> → CO + CO <sub>2</sub> + HC/ + H <sub>2</sub> O ✓	1	ALLOW any other correctly balanced equation with the same reactants and products ALLOW C:H;C/ for CH;CHC/
		(8)	Sodium hydrogencarbonate neutralises HC/ -/	1	Assume that 'it' refers to sodium hydrogencarbonate but DO NOT ALLOW other chemicals e.g. sodium  ALLOW NaHCO <sub>3</sub> is a base  ALLOW forms a salt or sodium chloride or NaCl  ALLOW equation to show formation of NaCl from NaHCO <sub>3</sub> and HCl even if not balanced.

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