



OXFORD CAMBRIDGE AND RSA EXAMINATIONS

Advanced Subsidiary GCE

**ACCOUNTING**

Unit F012: Accounting Applications

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**J Elliott**



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web jan 5 2016 explanation isotopes are atoms of the same element same atomic number i e same number of protons in the nucleus but different atomic masses different mass numbers due to different numbers of neutrons in the nucleus for example we may get an isotope of chlorine with 17 protons and 18 neutrons hence atomic mass 35

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neutrons massive nuclear particles of zero charge

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d more than one response is correct e they have different numbers of protons in the nucleus

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