

OCR A A-LEVEL ORGANIC CHEMISTRY

Complete Notes for
Modules 4 and 6

Alkenes 1

Chemistry

Alkenes are unsaturated hydrocarbons (contain a carbon-carbon double bond)



Alkenes have a high C=C to C-C bond ratio
C=C is 612 kJ mol⁻¹
C-C is 348 kJ mol⁻¹
Therefore a carbon-carbon double bond is stronger than a single bond
Therefore alkenes are more reactive than alkanes

Alkenes are unsaturated hydrocarbons (contain a carbon-carbon double bond)

Alkenes are more reactive than alkanes
Alkenes are more reactive than alkanes
Alkenes are more reactive than alkanes



Alkenes are more reactive than alkanes
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Alkenes are more reactive than alkanes

Polymers

Polymers are long chains of repeating units (monomers)

Polymers are long chains of repeating units (monomers)



Polymers are long chains of repeating units (monomers)



Alkenes 2

Alkenes are more reactive than alkanes
Alkenes are more reactive than alkanes
Alkenes are more reactive than alkanes



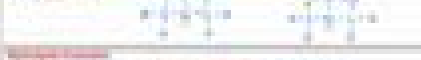
Alkenes are more reactive than alkanes
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Amines and Amides

Amines are compounds containing a nitrogen atom bonded to one or more hydrogen atoms



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Phenols

Phenols are compounds containing a hydroxyl group attached to a benzene ring



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Padhraic Smyth

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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