

Macbeth Questions – Act 4

1. How do the witches lull Macbeth into a false sense of security?
2. How does Macbeth feel after his second meeting with the witches? What things might reassure him, and what things might trouble him?
3. When Macbeth says (Act 4, scene 1) that “the flighty purpose never is o’ertook/Unless the deed go with it”, what does he mean in general, and in the particular situation in which he says it?
4. How does Lady Macduff feel about her husband’s leaving her alone? Why has he done this?
5. When Ross comes to see Lady Macduff, how does he behave and why?
6. Briefly explain Macduff’s description to Malcolm of the state of affairs in Scotland (Act 4, scene 3). What does the audience know which makes Macduff’s account even more painful?
7. Why might Malcolm be suspicious of Macduff? Does he know as much as the audience does about why Macbeth and Macduff are enemies?
8. Malcolm pretends to be even more evil than Macbeth. Why does he do this? What does he discover by doing so?
9. In Act 4, scene 3, there is an account of the miraculous healing powers of the English king - what is the purpose of this? What effect does it have on the audience?
10. When Macduff asks Ross about his family, Ross replies, “...they were well at peace when I did leave ‘em”. Explain why he gives this answer and what it might mean.

Objective Question Act 1 Macbeth

M Carnoy



Objective Question Act 1 Macbeth:

British and American Drama (English) For B.A. (Sem.3) According to NEP-2020 R. Bansal,2022-09-24 Contents 1 Drama Elements and Types 2 Literary Terms Drama 3 Othello By William Shakespeare 4 Macbeth By William Shakespeare 5 Arms And The Man By George Bernard Shaw 6 She Stoops To Conquer By Oliver Goldsmith 7 Look Back In Anger By John Osborne 8 Murder In The Cathedral By T S Eliot 9 A Street Car Named Desire By Tennessee Williams 10 The Glass Menagerie By Tennessee Williams 11 All My Sons By Arthur Miller Additional Information The author of this book is R Bansal

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web answer oxidation reductions reactions always have an electron transfer from the oxidized species to the reduced species when the oxidized species is separated from the

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web google classroom which of the following statements accurately describes the nernst equation shown below
 $E_{\text{cell}} = E^{\circ}_{\text{cell}} - \frac{0.0591}{n} \log Q$
choose 1 answer choose 1 answer choice a at equilibrium E_{cell} calculates to a value of 0 and the battery is

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web sep 27 2022 Cu^{2+} ions are lower than H^{+} ions in the electrochemical series therefore selectively discharged at the cathode
 Cu^{2+} ions have greater tendency to

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web ans it suggests that Zn is more reactive than hydrogen or that it is a more powerful reducing agent zinc will be oxidized to Zn^{2+} ions in a cell with a zinc electrode and a

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web $10 \log K$ we can plug in the value of G° on the left side of the equation even though G° is normally expressed as kJ mol^{-1} is expressed as J mol^{-1} so we can convert R or G° to match units let's plug in $300\,000 \text{ J}$ for G° to match R divide $300\,000$ by

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web sep 24 2020 ans 3.6 the cell in which the following reaction occurs $2\text{Fe}^{3+}(\text{aq}) + 2\text{I}^{-}(\text{aq}) \rightleftharpoons 2\text{Fe}^{2+}(\text{aq}) + \text{I}_2(\text{s})$ has $E_{\text{cell}} = 0.236 \text{ V}$ at 298 K calculate the standard gibbs

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web method 2 half reaction method step 1 determine oxidation numbers for each atom step 2 use oxidation numbers to determine what is oxidized and what is reduced step 3

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web reason when successive electrons are added to the orbitals in the same principal quantum level the shielding effect of inner core of electrons does not increase very much to

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web an electrochemical cell is any device that converts chemical energy into electrical energy or electrical energy into chemical energy there are three components that make up an

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