

SECOND MID TERM TEST - 2024

Standard XII

Reg.No.

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| 0 | 1 | 2 | 5 | 6 |
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CHEMISTRY

Time : 1.30 hrs

Part - I

Marks :

10 x 1 =

1. Choose the correct answer:

- A magnetic moment of 1.73 BM will be shown by one among the following
a) TiCl_4 b) $[\text{CoCl}_6]^{4-}$ c) $[\text{Cu}(\text{NH}_3)_4]^{2+}$ d) $[\text{Ni}(\text{CN})_4]^{2-}$
- Which type of isomerism is exhibited by $[\text{Pt}(\text{NH}_3)_2 \text{Cl}_2]$?
a) coordination isomerism b) linkage isomerism
c) optical isomerism d) geometrical isomerism
- The number of electrons that have a total charge of 9650 coulombs is
a) 6.22×10^{23} b) 6.022×10^{24} c) 6.022×10^{22} d) 6.022×10^{-34}
- Which of the following electrolytic solution has the least specific conductance?
a) 2 N b) 0.002 N c) 0.02 N d) 0.2 N

Physical Chemistry Second Midtterm Exam Questions

JR Anderson



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