

Physical Science: Element Worksheet

Name _____ Date _____

Mendeleev's Chart

- _____ 1. Mendeleev arranged the elements in his periodic table in order of:
- a. atomic number
 - b. increasing valence
 - c. mass
 - d. increasing valency.
- _____ 2. Mendeleev's decision to leave gaps in his periodic table was supported by the discovery of:
- a. electrons
 - b. protons
 - c. atoms
 - d. gallium.
- _____ 3. On Mendeleev's periodic table, elements with similar properties were grouped:
- a. In the same row
 - b. In groups of two
 - c. In the same column
 - d. Along diagonal lines that run from left to bottom-right.
- _____ 4. Which of the following is not a characteristic of metals?
- a. ductile
 - b. good conductors of electricity
 - c. typically solid at room temp.
 - d. Plastic.
- _____ 5. The elements from group 2 on the periodic table are called:
- a. Halogens
 - b. Alkaline Earth Metals
 - c. Alkalis
 - d. Noble Gases
- _____ 6. "What do elements in a group have similar chemical properties?"
- a. They have the same number of valence electrons
 - b. They have the same valence configuration of electrons
 - c. Their atomic masses are multiples of the least massive element of the group
 - d. They have the same number of total electrons.
- _____ 7. How many periods does the modern periodic table have?
- a. 5
 - b. 7
 - c. 10
 - d. 18
- _____ 8. "Which elements are all gases at room temp. (25° C)?"
- a. Group 1
 - b. Group 2
 - c. Group 7
 - d. Group 8
- _____ 9. The most reactive metals are the:
- a. Transition metals
 - b. Alkaline earths
 - c. Metalloids
 - d. Noble gases
- _____ 10. An element that is shiny and conducts electricity is likely to be a:
- a. Gas
 - b. metalloid
 - c. metal
 - d. nonmetal
- _____ 11. Moving up and down a vertical column on the periodic table would represent which of the following? (a) Period (b) Group (c) Family (d) A and B

Physical Science Element Answer Key

Beth Barber

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