

Physical properties of solutions

Even though “like dissolves like”, the presence of **solute molecules in the solvent**

disrupts the bulk solvent intermolecular forces to some extent.

This change in forces means that the **physical properties of solutions**,

like freezing and boiling points, **are slightly different than those of the pure solvent**.

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Applied Mechanics Reviews, 1974

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Journal of Chemical Education, 1929

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Nuclear Science Abstracts, 1974

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Solubilization in Surfactant Aggregates Sherril D. Christian, John F.

Scamehorn, 2020-08-27 This work covers topics ranging from fundamental studies of solubilization to practical technological applications of the phenomenon. It reviews the solubilization of organic materials into surfactant aggregates including micelles, vesicles, and admicelles. The book also details methods of measuring solubilization that utilize both classical and newer instrumental techniques. It is intended for physical surface, colloid, and surfactant chemists, chemical, environmental, and civil engineers, and upper level undergraduate and graduate students in these disciplines.

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