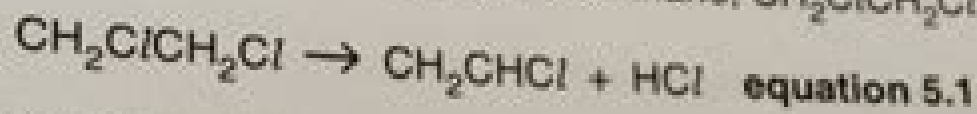


Chloroethene, CH_2CHCl , is made from 1,2-dichloroethane, $\text{CH}_2\text{ClCH}_2\text{Cl}$.



- (a) A chemical plant uses 19.80 tonnes of 1,2-dichloroethane to make 11.25 tonnes of chloroethene.

Calculate the percentage yield of chloroethene.

(1 tonne = $1.00 \times 10^6 \text{ g}$)

$$\% = \frac{\text{actual amount in mol}}{\text{Theoretical amount in mol}} \times 100$$

$$= \frac{\frac{11.25 \times 10^6}{62.5}}{\frac{19.8 \times 10^6}{98}} \times 100 =$$

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Michael Rajnik



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Table of Contents Ocr Chemistry Past Paper F322 June 2014

1. Understanding the eBook Ocr Chemistry Past Paper F322 June 2014
 - The Rise of Digital Reading Ocr Chemistry Past Paper F322 June 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Chemistry Past Paper F322 June 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Chemistry Past Paper F322 June 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Chemistry Past Paper F322 June 2014
 - Personalized Recommendations

- Ocr Chemistry Past Paper F322 June 2014 User Reviews and Ratings
- Ocr Chemistry Past Paper F322 June 2014 and Bestseller Lists
- 5. Accessing Ocr Chemistry Past Paper F322 June 2014 Free and Paid eBooks
 - Ocr Chemistry Past Paper F322 June 2014 Public Domain eBooks
 - Ocr Chemistry Past Paper F322 June 2014 eBook Subscription Services
 - Ocr Chemistry Past Paper F322 June 2014 Budget-Friendly Options
- 6. Navigating Ocr Chemistry Past Paper F322 June 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr Chemistry Past Paper F322 June 2014 Compatibility with Devices
 - Ocr Chemistry Past Paper F322 June 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Chemistry Past Paper F322 June 2014
 - Highlighting and Note-Taking Ocr Chemistry Past Paper F322 June 2014
 - Interactive Elements Ocr Chemistry Past Paper F322 June 2014
- 8. Staying Engaged with Ocr Chemistry Past Paper F322 June 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Chemistry Past Paper F322 June 2014
- 9. Balancing eBooks and Physical Books Ocr Chemistry Past Paper F322 June 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Chemistry Past Paper F322 June 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr Chemistry Past Paper F322 June 2014
 - Setting Reading Goals Ocr Chemistry Past Paper F322 June 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr Chemistry Past Paper F322 June 2014
 - Fact-Checking eBook Content of Ocr Chemistry Past Paper F322 June 2014

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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