

Answer all the questions.

5. Methanoic acid, HCOOH , is a carboxylic acid that is present in some stings.

(a) What is the formula of the functional group that is responsible for the characteristic properties of carboxylic acids?

[2]

(b) Methanoic acid is usually referred to simply as formic acid.

Formic acid is made of carbon, hydrogen, oxygen and is insoluble in water.

Carboxylic acids react with carbonates in a similar way to other acids such as hydrochloric acid.



(c) Complete and balance the symbol equation for the reaction between calcium carbonate and methanoic acid.



[2]

(d) Calcium carbonate is insoluble so it does not react with water.

When calcium carbonate is allowed to react with methanoic acid, calcium methanoate forms.

The reaction with methanoic acid removes the calcium carbonate in limestone. Suggest a property of calcium methanoate that can explain why this happens.

[1]

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Table of Contents Ocr 21st Century Science Chemistry C7 Specimen

1. Understanding the eBook Ocr 21st Century Science Chemistry C7 Specimen
 - The Rise of Digital Reading Ocr 21st Century Science Chemistry C7 Specimen
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr 21st Century Science Chemistry C7 Specimen
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr 21st Century Science Chemistry C7 Specimen
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr 21st Century Science Chemistry C7 Specimen
 - Personalized Recommendations
 - Ocr 21st Century Science Chemistry C7 Specimen User Reviews and Ratings
 - Ocr 21st Century Science Chemistry C7 Specimen and Bestseller Lists

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 - ePub, PDF, MOBI, and More
 - Ocr 21st Century Science Chemistry C7 Specimen Compatibility with Devices
 - Ocr 21st Century Science Chemistry C7 Specimen Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr 21st Century Science Chemistry C7 Specimen
 - Highlighting and Note-Taking Ocr 21st Century Science Chemistry C7 Specimen
 - Interactive Elements Ocr 21st Century Science Chemistry C7 Specimen
8. Staying Engaged with Ocr 21st Century Science Chemistry C7 Specimen
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr 21st Century Science Chemistry C7 Specimen
9. Balancing eBooks and Physical Books Ocr 21st Century Science Chemistry C7 Specimen
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr 21st Century Science Chemistry C7 Specimen
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr 21st Century Science Chemistry C7 Specimen
 - Setting Reading Goals Ocr 21st Century Science Chemistry C7 Specimen
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr 21st Century Science Chemistry C7 Specimen
 - Fact-Checking eBook Content of Ocr 21st Century Science Chemistry C7 Specimen
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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