

A2 Unit F334: Chemistry of Materials

Topic 3: What's in a medicine? (20%)

A study of medicines such as aspirin, their development, chemistry and synthesis, illustrating some of the features of the pharmaceutical industry. The chemical ideas in this module are:

- Physical properties such as colour, odour and solubility
- Acid-base reactions
- Polymer manufacture and testing
- IR spectroscopy and mass spectrometry

Topic	Alt code	Number	Assessable learning outcomes	Number of questions	ICP A2 Revision guide (Page number)	Chemical changes (Page number)	Chemical ideas (Page number)
Equilibria		10401	Describe acids in terms of the Brønsted-Lowry theory as proton donors, and bases as proton acceptors.	1	6		100
		10402	Identify the proton donor and proton acceptor in an acid-base reaction.	1	6		100
Bonding and structure	1110	10403	Draw and interpret simple electron-dot-and-cross diagrams to show how atoms (bond through one, covalent and three covalent bonds).	1	10		101-103
	1110	10404	Describe a simple model of metallic bonding.		1		4
	1110	10405	Use the electron-dot-repulsion principle to predict and explain the shapes of simple molecules (such as CH ₄ , NH ₃ , H ₂ O and SF ₆) and ions (such as BF ₄ ⁻) with up to six outer pairs of electrons. (Any combination of bonding pairs and lone pairs (no treatment of hybridisation or molecular orbital) is accepted).	1	7		101-103

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AN Whitehead



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