

Name: _____

Branches of Chemistry Multiple Choice

Circle the correct answer.

1. This branch of Chemistry studies the properties and changes of matter and their relation to energy.
a. organic c. biochemistry e. physical
b. inorganic d. analytical f. theoretical
2. This branch of Chemistry studies most carbon-containing compounds.
a. organic c. biochemistry e. physical
b. inorganic d. analytical f. theoretical
3. This branch of Chemistry identifies the composition of materials.
a. organic c. biochemistry e. physical
b. inorganic d. analytical f. theoretical
4. This branch of Chemistry studies the substances and processes in living things.
a. organic c. biochemistry e. physical
b. inorganic d. analytical f. theoretical
5. This branch of Chemistry studies non-carbon molecules.
a. organic c. biochemistry e. physical
b. inorganic d. analytical f. theoretical
6. This branch of Chemistry uses mathematics and computers to design and predict new substances.
a. organic c. biochemistry e. physical
b. inorganic d. analytical f. theoretical

Physical Chemistry Multiple Choice

Alyn G. McFarland, Nora Henry



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Physical Chemistry: A Level Multiple Choice Questions Margaret Marion Langley Oblitas, 1970-01-01 Multiple Choice Questions in A-level Chemistry B. P. Bourne, 1969 *Multiple Choice Questions in Physical Chemistry for Undergraduate Students* Dr Kusum Baser, 2022-08-25 Objective questions are very much useful for practice of particular topic Multiple choice questions are important to practice any topic to solve objective questions very deep study and knowledge of specific topic is needed This book helps the students to do practice in such a way so they can do preparation for competitive exams This book covers important topics of physical chemistry for undergraduate students Working Method Approach for Introductory Physical Chemistry Calculations Brian J Hathaway, Clair Murphy, Brian Murphy, 2019-05-16 A Working Method Approach for Introductory Physical Chemistry Calculations is a concise inexpensive introduction to first year chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem

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