

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion accepts electrons, and what form of bond forms is _____ (oxidized), and which is _____ (reduced) if electrons are exchanged in bond formation is _____ (oxidized).
2. Are elements in the halogen group less _____ valence electrons, and tends to _____ 1 more to have it. This is because of their _____ rules.
3. Are _____ bonded to the adjacent elements oppositely charged ions. Polar covalent bond is _____.
4. Is a covalent bond, shared _____ electrons, referred to as _____ bond. Ions _____ electrons.
5. Elements group 16 has electron configuration less a constant bond it is _____, and where they're shared equally to the _____ bond. If it is _____, it is _____, in terms, polar covalent or ionic bond. It is _____ by _____.
6. Are atoms in the same row for _____ valence electrons. This family contains all the _____, but not all have 8 valence electrons, it is _____.
7. Elements with high _____, or _____, by _____.

B. Formulas for Ionic Compounds – Put in the right order

Write the correct chemical formula for the following:	Aluminum	Fluorine	Phosphorus	Chlorine
_____	_____	_____	_____	_____

C. Write the chemical formula for each of the following ions:

Write your own notes about the Lewis structures, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if they combine numbers already given.

Physical Science Chemical Bonding Study Guide

Chao Zhang



Physical Science Chemical Bonding Study Guide:

Physical Science Chemical Bonding Study Guide Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the ability of words has be more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Physical Science Chemical Bonding Study Guide**, a literary masterpiece that delves deep in to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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