

Chemical and Physical Properties

State whether each of the following is a physical property or a chemical property, and type of property.

Description	Physical/ Chemical	Type
1. The sulphur is yellow.	Physical	Colour
2. Iron reacts with sulphur to give heat and a flame.	Chemical	Combustibility; Reaction
3. Baking soda with acid produces bubbles of gas.	Chemical	Reaction with acid
4. Metal can be rolled into flat sheets.	Physical	Malleability
5. Oxygen is colourless, odourless and tasteless.	Physical	Colour, odour, taste
6. Nitrogen dioxide gas has a choking smell.	Physical	Odour
7. Helium is less dense than air, so a helium balloon floats.	Physical	Density
8. Wax candle burns in oxygen.	Chemical	Combustibility
9. An apple rots due to fungi.	Chemical	Reaction
10. Mercury metal is a liquid.	Physical	State
11. Iodine gas is purple.	Physical	Colour
12. Hydrofluoric acid is poisonous.	Chemical	Reaction
13. Gold is a shiny metal.	Physical	Lustre
14. Cesium is the only other gold-coloured metal.	Physical	Colour
15. Potassium burns with a purple flame to make a white powder.	Chemical	Combustibility
16. Copper nitrate is a blue crystal.	Physical	Colour, crystal form
17. Two colourless solutions mix to give a yellow solid precipitate.	Chemical	Reaction
18. Copper can be stretched into a thin wire.	Physical	Ductility
19. Copper metal reacts with nitric acid to make a brown gas.	Chemical	Reaction with acid
20. Salt, sodium chloride, is a white crystal that melts at 801 °C.	Physical	Colour, crystal form, melting point

Physical Properties Of Matter Answer Key

Tobias Bleicker



Physical Properties Of Matter Answer Key:

Properties of Matter: Physical Properties of Matter Gr. 5-8 George Graybill,2015-09-01 This is the chapter slice Physical Properties of Matter from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

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Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **Properties of Matter:**

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Academic Vocabulary Level 4--The Scientific Method Stephanie Paris, 2014-01-01 This lesson integrates academic vocabulary instruction into content area lessons Two easy to implement strategies for teaching academic vocabulary are integrated within the step by step standards based science lesson

Understanding General Chemistry Atef Korchef, 2022-03-07 Understanding General Chemistry details the fundamentals of general chemistry through a wide range of topics relating the structure of atoms and molecules to the properties of matter Written in an easy to understand format with helpful pedagogy to fuel learning the book features main objectives at the beginning of each chapter get smart sections and check your reading section at the end of each chapter The text is filled with examples and practices that illustrate the concepts at hand In addition a summary and extensive MCQs exercises and problems with the corresponding answers and explanations are readily available Additional features include Alerts students to common mistakes and explains in simple ways and clear applications how to avoid these mistakes Offers answers and comments alongside sample problems enabling students to self evaluate their skill level Includes powerful methods easy steps simple and accurate interpretations and engaging applications to help students understand complex principles Provides a bridge to more complex topics such as solid state chemistry organometallic chemistry chemistry of main group elements inorganic chemistry and physical chemistry This introductory

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