

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) Which statement concerning Arrhenius acid-base theory is **not** correct? 1) _____
A) Acid-base reactions must take place in aqueous solution.
B) An Arrhenius acid produces hydrogen ions in water solution.
C) An Arrhenius base produces hydroxide ions in water solution.
D) A neutralization reaction produces water plus a salt.
E) none of the above
- 2) The H_3O^+ ion is called the _____ ion. 2) _____
A) protium
B) hydronium
C) hydrogen
D) hydroxide
E) water
- 3) When acids and bases react the product other than water is a 3) _____
A) hydroxide ion.
B) hydrogen ion.
C) hydronium ion.
D) salt.
E) metal.
- 4) Which of the following compounds is a salt? 4) _____
A) NaOH B) KNO_3 C) $\text{C}_6\text{H}_{12}\text{O}_6$ D) H_2SO_4 E) HBr
- 5) Which of the following compounds is a salt? 5) _____
A) $\text{CH}_3\text{CO}_2\text{H}$
B) NH_4NO_3
C) C_6H_6
D) NH_3
E) $\text{Al}(\text{OH})_3$
- 6) Which compound is manufactured in larger quantities in the U.S. than any other industrial chemical? 6) _____
A) HNO_3 B) H_2SO_4 C) HCl D) H_3PO_4 E) NaOH

Parvin Carter Chemistry

William Simon



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30% of researchers worldwide are women Long standing biases and gender stereotypes are discouraging girls and women away from science related fields and STEM research in particular Chemistry is no exception to this Science and gender equality are however essential to ensure sustainable development as highlighted by UNESCO In order to change traditional mindsets gender equality must be promoted stereotypes defeated and girls and women should be encouraged to pursue STEM careers

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