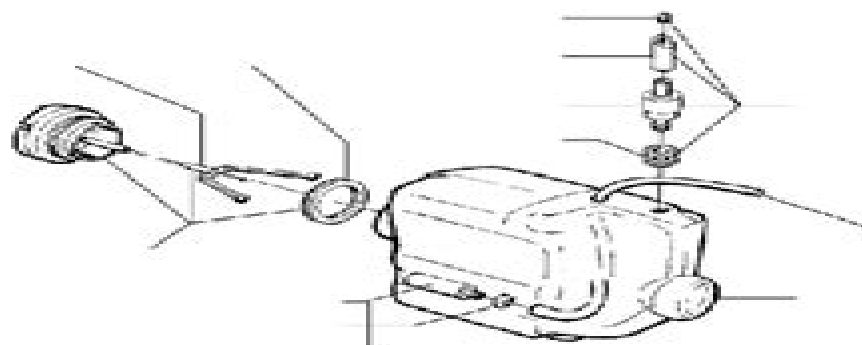
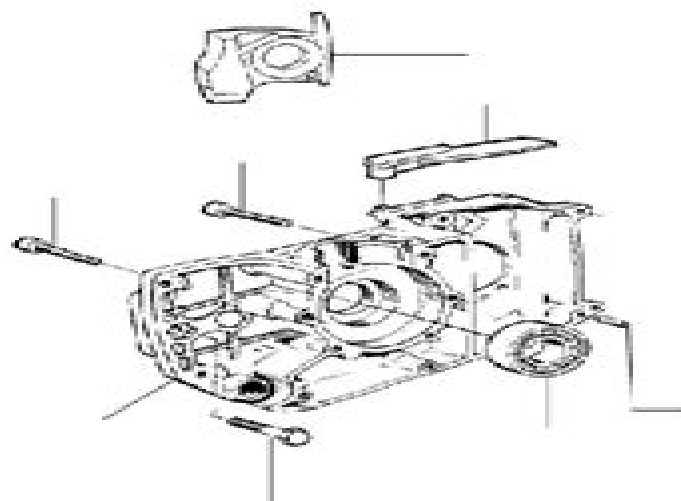
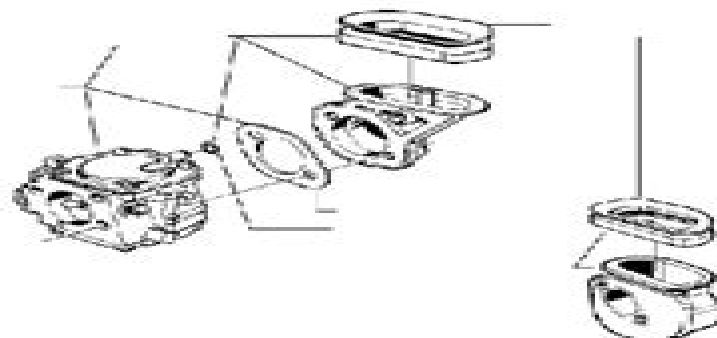
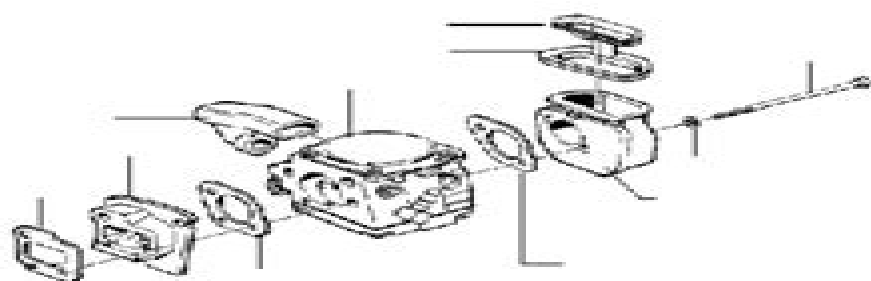


# Partner K700 Active 2 Manual

**Peipei Pang**



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because after its combustion a new substance is formed and cannot be turned back into  
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web q5 identify the reducing agent in the following reactions a  $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$  b  $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  c  $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$  d  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  answer a here ammonia  $\text{NH}_3$  is the reducing agent b here water  $\text{H}_2\text{O}$  is the reducing agent