

Use the partial periodic table to answer questions 4 -7.

⁶
C
Carbon
12.0107

— Atomic number
 — Symbol
 — Name
 — Average Atomic Mass

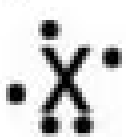
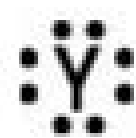
Partial Periodic Table of the Elements

1A	1 H Hydrogen 1.00794	2A	2 He Helium 4.0026
1A	3 Li Lithium 6.941	2A	4 Be Beryllium 9.0122
1A	11 Na Sodium 22.98976928	2A	12 Mg Magnesium 24.304
1A	19 K Potassium 39.0983	2A	20 Ca Calcium 40.078

3A	13 B Boron 10.811	4A	14 C Carbon 12.0107	5A	15 N Nitrogen 14.0064	6A	16 O Oxygen 15.9994	7A	17 F Fluorine 18.9984	8A	18 Ne Neon 20.1797
3A	13 Al Aluminum 26.9815385	4A	14 Si Silicon 28.0855	5A	15 P Phosphorus 30.973761	6A	16 S Sulfur 32.06	7A	17 Cl Chlorine 35.453	8A	18 Ar Argon 39.948

Question 5

Suppose scientists discovered four new elements (W, X, Y, Z) while studying rock and soil samples brought back from a Mars mission. Which Lewis dot structure represents an element that should be placed in column VIIA (17) of the periodic table?

- A.  B.  C.  D. 

Partial Periodic Answer Physical Science Lab

Brendan G. Carr



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