

- (d) The student found that  $22.50\text{ cm}^3$  of  $0.0800\text{ mol dm}^{-3}$   $\text{Na}_2\text{B}_4\text{O}_7$  reacted with  $25.00\text{ cm}^3$  of dilute hydrochloric acid.



- (i) Calculate the amount, in mol, of  $\text{Na}_2\text{B}_4\text{O}_7$  used.

amount = ..... mol [1]

- (ii) Calculate the amount, in mol, of  $\text{HCl}$  used.

amount = ..... mol [1]

- (iii) Calculate the concentration, in  $\text{mol dm}^{-3}$ , of the  $\text{HCl}$ .

concentration = .....  $\text{mol dm}^{-3}$  [1]

[Total: 12]

**END OF QUESTION PAPER**

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