F324-unofficial mark scheme (in progress)

Marks accounted for: 60/60

- Give the systematic name of the alcohol that is formed by this hydrolysis. (1)
  - Propan(e)-1,2,3-triol
- What are the other monomers that make up this polymer? (3)

  (Hydrolysed by hot NaOH)

  - Two different carboxylic acids with COO⁻(Na⁺) - The monomers have COOH as it's not referring to the hyrolysed products but the monomer units of the polymer I'm pretty sure that the question said: "What are the other products of this hydrolysis" or something along those lines
  - One salt had a c=c double bond
- Why are people who drink a lot of goat's milk likely to suffer from CHD? (2)
  - Fatty acids build up in the arteries and lead to CHD?
  - Has trans isomer due to c=c double bond fatty acid linked to LDL's and hence Atherosclerosis
  - *Trans*-fatty acid in triglyceride of milk increases 'bad' cholesterol levels in blood, resulting in increased risk of coronary heart disease and stroke (straight from the spec)
  - trans fatty acids can pack more tightly and hence have higher melting points than cis fatty acids. --> so more likely to deposit on artery walls
  - The other two molecules were fully saturated fatty acids which would also lead to the formation of atheroma

- State how 4-aminophenol can behave as a base. (1)
- The lone pair on the Nitrogen atom in the amine group can accept a proton, forming a dative covalent bond with it to give NH3+.
- NH2 + H+ ---> NH3+

- 4-Aminophenol is produced by the reduction of 4-nitrophenol. Write an equation to show the production of 4-aminophenol from 4-nitrophenol. Use [H] to represent the reducing agent. (1)
- 4-Nitrophenol + 6[H] = 4 Aminophenol + 2H20

- 4-nitrophenol can be produced from 4-bromophenol. (4)
  - Complete the mechanism for this reaction.
  - Use "NO₂ as the electrophile. Include any intermediate and the products.
  - In the mechanism for this reaction, NO₂ substitutes for Br on the ring.

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