



**PHYSICAL SCIENCES (CHEMISTRY)  
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

**TIME: 1 HOUR**

**VERIFICATION OF BOYLE'S LAW  
INDIVIDUAL WORK (should be done under strict supervision).**

**1. Write a scientific report which must include the following:**

**1.1 Aim**

**(1)**

**1.2 Apparatus and method**

**(5)**

**2. Observations and Recording of results** (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate  $pV$  for each set of results and record your answers in the table.

Volume (V) $\text{cm}^3$	Pressure (p) kPa		$pV$ (J)

**(6)**

**3. Analysis of the results:**

**3.1 Identify the following variables for the experiment:**

- Independent
- Dependent
- Controlled

**(3)**

**3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus  $\frac{1}{p}$ .**

**(18)**

**4. Conclusion**

**4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.**

**(1)**

**4.2 Write down a conclusion for this experiment.**

**(3)**

**5**

**Questions and discussion:**

**5.1 Explain why you have to wait for a few minutes before you can take readings.**

**(2)**

**5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.**

**(4)**

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**J Spring**



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## **Table of Contents Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga**

1. Understanding the eBook Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - The Rise of Digital Reading Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - Personalized Recommendations
  - Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga User Reviews and Ratings
  - Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga and Bestseller Lists
5. Accessing Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga Free and Paid eBooks

- Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga Public Domain eBooks
  - Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga eBook Subscription Services
  - Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga Budget-Friendly Options
6. Navigating Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga eBook Formats
- ePub, PDF, MOBI, and More
  - Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga Compatibility with Devices
  - Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - Highlighting and Note-Taking Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - Interactive Elements Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
8. Staying Engaged with Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
9. Balancing eBooks and Physical Books Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
- Setting Reading Goals Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
- Fact-Checking eBook Content of Physical Science Memo Paper 1 2013 Grade 11 Mpumalanga
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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