

PHYSICAL CHEMISTRY

(Complete Syllabus)

SECTION - A

Straight Objective Type

This section contains 220 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

- The mass of oxygen required to produce enough CO which completely reduces x gm of Fe_2O_3 . Suppose molecular weight of $\text{O}_2 = y$, molecular weight of $\text{Fe}_2\text{O}_3 = z$
(1) $\frac{2xz}{3y}$ (2) $\frac{3xy}{2z}$
(3) $\frac{2y}{3xz}$ (4) $\frac{3z}{2xy}$
- The number of lone pair of electrons present in 2.56 gm of orthorhombic sulphur will be
(1) $0.02 N_0$ (2) $0.16 N_0$
(3) $0.32 N_0$ (4) $0.08 N_0$
- A 10.0 g of sample of mixture of calcium chloride and sodium chloride is treated with Na_2CO_3 to precipitate the calcium as calcium carbonate. This CaCO_3 is heated to convert all calcium to CaO and final mass of CaO is 1.62 gms. The % by mass of CaCl_2 in the original mixture is
(1) 15.2%
(2) 32.1%
(3) 21.8%
(4) 11.07%
- 1000 ml of oxygen at STP were passed through an ozoniser and the resulting volume was 888 ml at STP. This quantity of ozonised oxygen is passed through excess of KI solution. The weight of I_2 liberated is
(1) 25.4 g (2) 12.7 g
(3) 2.54 g (4) 1.27 g
- x_1 ml solution having normality 1 is diluted to get a solution of normality y_2 . What would be volume required for dilution?
(1) $x_1 \left(1 - \frac{y_1}{y_2}\right)$ (2) $x_1 \left(\frac{y_2}{y_1} - 1\right)$
(3) $x_1 \left(\frac{y_1}{y_2} - 1\right)$ (4) $x_1 \left(1 + \frac{y_1}{y_2}\right)$
- 1.625 g of an interhalogen compound ICl_x was subjected to electric discharge which split it into constituents and brought to STP. It occupied 112 ml. The interhalogen compound is
(1) ICl (2) ICl_3
(3) ICl_5 (4) ICl_7
- If 12.25 g of a sample of KClO_3 of 75% purity and rest vapourisable impurities is strongly heated to give KCl , KClO_4 and 0.32 gm O_2 . The mass of residue obtained is
(1) 1.769 g (2) 7.096 g
(3) 5.327 g (4) 8.85 g
- A 2 g sample of xenon reacts with fluorine. The mass of compound produced is 3.158 g. The empirical formula of the compound is
(1) XeF_2 (2) XeF_4
(3) XeF_6 (4) XeF_8
- 1 g of carbonate of metal was dissolved in 25 ml of 1 N HCl. The resulting liquid required 5 ml of 1 N NaOH for neutralisation. The equivalent weight

Physical Chemistry Exam Questions

IM Harris



Physical Chemistry Exam Questions:

Physical Chemistry Kenneth S Schmitz, 2018-06-14 The advancements in society are intertwined with the advancements in science To understand how changes in society occurred and will continue to change one has to have a basic understanding of the laws of physics and chemistry Physical Chemistry Multidisciplinary Applications in Society examines how the laws of physics and chemistry physical chemistry explain the dynamic nature of the Universe and events on Earth and how these events affect the evolution of society multidisciplinary applications The ordering of the chapters reflects the natural flow of events in an evolving Universe Philosophy of Science the basis of the view that natural events have natural causes Cosmology the origin of everything from the Big Bang to the current state of the Universe Geoscience the physics and chemistry behind the evolution of the planet Earth from its birth to the present Life Science the molecules and mechanisms of life on Earth Ecology the interdependence of all components within the Ecosphere and the Universe Information Content emphasis on how words and phrases and framing of issues affect opinions reliability of sources and the limitations of knowledge Addresses the four Ws of science Why scientists believe Nature works the way it does Who helped develop the fields of science What theories of natural processes tell us about the nature of Nature and Where our scientific knowledge is taking us into the future Gives a historical review of the evolution of science and the accompanying changes in the philosophy of how science views the nature of the Universe Explores the physics and chemistry of Nature with minimal reliance on mathematics Examines the structure and dynamics of the Universe and our Home Planet Earth Provides a detailed analysis of how humans as members of the Ecosphere have influenced and are continuing to influence the dynamics of events on the paludarium called Earth Presents underlying science of current political issues that shape the future of humankind Emphasizes how words and phrases and framing of issues can influence the opinions of members of society Makes extensive use of metaphors and everyday experiences to illustrate principles in science and social interactions Exam Survival Guide: Physical Chemistry Jochen Vogt, 2017-02-24 A text and exercise book for physical chemistry students This book deals with the fundamental aspects of physical chemistry taught at the undergraduate level in chemistry and the engineering sciences in a compact and practice oriented form Numerous problems and detailed solutions offer the possibility of an in depth reflection of topics like chemical thermodynamics and kinetics atomic structure and spectroscopy Every chapter starts with a recapitulation of important background information before leading over to representative exercises and problems Detailed descriptions systematically present and explain the solutions to the problems so that readers can carefully check their own solutions and get clear cut introductions on how to approach similar problems systematically The book addresses students at the upper undergraduate level as well as tutors and teachers It is a rich source of exercises for exam preparation and can be used alongside classical textbooks Furthermore it can serve teachers and tutors for the conception of their lessons Its well thought through presentation structure and design make the book appeal to everybody who wants to succeed with the

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Student Assessment in Higher Education Kevin Cox,Bradford Imrie,Allen Miller,2014-05-12 This text provides higher education teachers with an overview of the many approaches to setting marking and reviewing coursework assignments tests and examinations used in programmes for certificates diplomas first degrees or higher degrees It discusses the influence of each on students

Understanding Physical Chemistry ,1964 *Mosby's Review Questions for the NCLEX-RN Exam - E-Book* Patricia M. Nugent,Judith S. Green,Barbara A. Vitale,Phyllis K. Pelikan,2010-11-02 This title includes additional digital media when purchased in print format For this digital book edition media content is not included Prepare for exam success with Mosby's Review Questions for the NCLEX RN Examination Over 5 000 exam style practice questions help you assess your strengths and weaknesses develop test taking skills and reduce your test anxiety Written by testing experts Patricia M Nugent Phyllis K Pelikan Judith S Green and Barbara A Vitale this book makes review easy by organizing material into the core clinical areas of medical surgical nursing mental health maternity and pediatrics Rationales are provided for both correct and incorrect answers and alternate item format questions ensure that you're prepared for the latest version of the exam Convenient organization by core clinical area body system and disorders makes it easy for students to select the practice questions they prefer More than 3 000 questions in the book Rationales for both correct and incorrect answers explain the reasoning behind each answer option Alternate item format questions include fill in the blank multiple response drag and drop prioritizing chart exhibit and hot spot figure illustration enhance students critical thinking skills Three practice modes study quiz and exam Coverage of new content on the 2010 NCLEX RN test plan prepares your students for the exam with the most up to date information An increase to over 300 alternate item format questions provides additional practice with these important critical thinking questions 12 chart exhibit alternate item format questions introduce students

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Table of Contents Physical Chemistry Exam Questions

1. Understanding the eBook Physical Chemistry Exam Questions
 - The Rise of Digital Reading Physical Chemistry Exam Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Chemistry Exam Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Physical Chemistry Exam Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Chemistry Exam Questions
 - Personalized Recommendations

- Physical Chemistry Exam Questions User Reviews and Ratings
- Physical Chemistry Exam Questions and Bestseller Lists
- 5. Accessing Physical Chemistry Exam Questions Free and Paid eBooks
 - Physical Chemistry Exam Questions Public Domain eBooks
 - Physical Chemistry Exam Questions eBook Subscription Services
 - Physical Chemistry Exam Questions Budget-Friendly Options
- 6. Navigating Physical Chemistry Exam Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Chemistry Exam Questions Compatibility with Devices
 - Physical Chemistry Exam Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Chemistry Exam Questions
 - Highlighting and Note-Taking Physical Chemistry Exam Questions
 - Interactive Elements Physical Chemistry Exam Questions
- 8. Staying Engaged with Physical Chemistry Exam Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Chemistry Exam Questions
- 9. Balancing eBooks and Physical Books Physical Chemistry Exam Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Chemistry Exam Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Chemistry Exam Questions
 - Setting Reading Goals Physical Chemistry Exam Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Chemistry Exam Questions
 - Fact-Checking eBook Content of Physical Chemistry Exam Questions

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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