Physical Science Study Guide Chapter 4

Name:	

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- Which of the following scientists inferred that an atom's positive charge must be clustered in the nucleus?
 a. Niels Bohr b. John Dalton c. Ernest Rutherford d. J.J. Thomson
- Which particles in atoms have a negative electric charge?
 a. electrons b. protons c. neutrons d. nuclei
- Mendeleev created the first periodic table by arranging elements in order of
 a decreasing atomic mass. b. increasing atomic mass. c. increasing atomic number. d. increasing melting points
 and densities.
- 4. What prediction did Mendeleev make that came true less than 20 years later?
 a. He predicted the atomic numbers of unknown elements. b. He predicted that a total of 112 elements would be discovered. c. He said that three new elements would be discovered, and he described their properties. d. He said that the periodic table would be developed into 18 families.
- The elements in a column of the periodic table

 a. have similar properties.
 b. are in the same period.
 c. have the same atomic mass.
 d. have very similar chemical symbols.
- Most metals are NOT

 a. ductile.
 b. good conductors of heat and electricity.
 c. liquid at room temperature.
 d. malleable.
- In general, which of the following statements about metals is true?
 a. Metals need to be stored in sealed containers for safety.
 b. Metals show a wide range of chemical properties.
 c. Metals are highly reactive substances.
 d. Metals do not react with oxygen.
- Which group contains the most elements?
 a. semimetals b. nonmetals c. metals d. transition elements
- In the periodic table, the most reactive metals are found

 in Group 1, the first column on the left.
 in Period 1, the first row across the top.
 in Groups 13 through 16 in the center.
 in Periods 6 and 7 at the bottom.
- 10. Which of the following statements about transition metals is true?
 a. They are never found uncombined in nature. b. They include familiar metals such as gold, silver, copper, and nickel. c. They are so soft that they can be cut with an ordinary knife. d. They are the most reactive of all the types of metals.
- Which property of bromine could you NOT predict based on the fact that it is a nonmetal in the halogen family?
 a. highly reactive b. poor conductor of electricity c. liquid at room temperature d. poor conductor of heat
- The elements that do not ordinarily form compounds are a. elements in the carbon family. b. metals. c. halogens. d. inert gases.
- Fluorine, chlorine, bromine, and iodine are part of a family called a. inert gases. b. semimetals. c. halogens. d. alkali metals.
- 14. Which group of elements shares characteristics with both metals and nonmetals?
 a. salts b. semimetals c. halogens d. lanthanides
- 15. In 1896, the French scientist Henri Becquerel discovered

Physical Science Chapter 4 Study Guide Answers

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Physical Science Chapter 4 Study Guide Answers:

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Student Study.

Guide Peter L. Kresan, Reed Mencke, Frank Press, Raymond Siever, 2003-09-25. This reconceptualization of the text Understanding Earth reflects the fundamental changes in the field of physical geology over the past several years.

Succeeding in Your School Entrance Exams: A Comprehensive Guide Pasquale De Marco, 2025-04-23 This comprehensive guidebook is the ultimate resource for students preparing for the Scholastic Testing Service High School Placement Test HSPT and the Cooperative Entrance Examination COOP Whether you re aiming for admission to a prestigious private or parochial high school or seeking to strengthen your academic foundation for future endeavors this book will equip you with the knowledge strategies and confidence you need to excel Unparalleled Content Review Our team of expert educators has meticulously analyzed the content and structure of both exams to provide you with a thorough understanding of what to expect on test day From fundamental math concepts to intricate grammar rules from scientific principles to historical events we leave no stone unturned in our quest to equip you with the knowledge and skills necessary for triumph Proven Test Taking Strategies Beyond content mastery we recognize the importance of effective test taking strategies This book is replete with proven techniques to help you manage your time wisely tackle various question formats with confidence and overcome the anxiety that often accompanies standardized testing We provide invaluable insights into the minds of test writers revealing their patterns and tendencies so that you can anticipate their moves and outsmart the exams Abundant Practice Opportunities Practice makes perfect and this book provides an abundance of practice questions carefully designed to mirror the actual HSPT and COOP exams in terms of difficulty and format These practice questions are not mere repetitions they are opportunities for you to apply your knowledge identify your strengths and weaknesses and refine your

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- 5. In chapter 4, this book will scrutinize the relevance of Physical Science Chapter 4 Study Guide Answers in specific contexts. The fourth chapter will explore how Physical Science Chapter 4 Study Guide Answers is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, this book will draw a conclusion about Physical Science Chapter 4 Study Guide Answers. The final chapter will summarize the key points that have been discussed throughout the book.
 - The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Physical Science Chapter 4 Study Guide Answers.

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