

# PHYSICAL SCIENCES P.2

# MEMO-2024



basic education

Department:  
Basic Education  
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL  
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

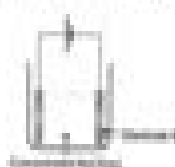
The inorganic reagents shown below are available for reactions IV and Y.

Br <sub>2</sub>	H <sub>2</sub> SO <sub>4</sub> (conc.)	NaOH(aq.)	HBr	H <sub>2</sub>
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reaction in the above diagram that is a redox reaction.



4.8 Write the half-cell reaction for the anode in the above cell.

4.9 Write the overall cell reaction for the above cell.

4.10 Write down the name of the reaction in the above diagram.

4.11 Write down the name of the reaction in the above diagram.



4.12 Write down the name of the reaction in the above diagram.

4.13 Write down the name of the reaction in the above diagram.

4.14 Write down the name of the reaction in the above diagram.

4.15 Write down the name of the reaction in the above diagram.

# Physical Science Grade November 2013 Memo

**Sebastian Brünink**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name Sebastian Brünink.

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## **Table of Contents Physical Science Grade November 2013 Memo**

1. Understanding the eBook Physical Science Grade November 2013 Memo
  - The Rise of Digital Reading Physical Science Grade November 2013 Memo
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade November 2013 Memo
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Grade November 2013 Memo
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade November 2013 Memo
  - Personalized Recommendations
  - Physical Science Grade November 2013 Memo User Reviews and Ratings
  - Physical Science Grade November 2013 Memo and Bestseller Lists
5. Accessing Physical Science Grade November 2013 Memo Free and Paid eBooks
  - Physical Science Grade November 2013 Memo Public Domain eBooks
  - Physical Science Grade November 2013 Memo eBook Subscription Services
  - Physical Science Grade November 2013 Memo Budget-Friendly Options

6. Navigating Physical Science Grade November 2013 Memo eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Grade November 2013 Memo Compatibility with Devices
  - Physical Science Grade November 2013 Memo Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Grade November 2013 Memo
  - Highlighting and Note-Taking Physical Science Grade November 2013 Memo
  - Interactive Elements Physical Science Grade November 2013 Memo
8. Staying Engaged with Physical Science Grade November 2013 Memo
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Grade November 2013 Memo
9. Balancing eBooks and Physical Books Physical Science Grade November 2013 Memo
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Grade November 2013 Memo
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Grade November 2013 Memo
  - Setting Reading Goals Physical Science Grade November 2013 Memo
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Grade November 2013 Memo
  - Fact-Checking eBook Content of Physical Science Grade November 2013 Memo
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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