

**CA - CPT – June 2014 Question Paper (Based on Memory)**

**1<sup>st</sup> Session**

**Marks: 100 Marks**

**Time: 2 hrs.**

**Part A – Fundamentals of Accounting**

1. Which of the following is wrong?  
a) All real and personal accounts are transferred to balance sheet  
b) Nominal accounts are transferred to P & L account  
c) Each account is opened separately in ledger  
d) Rent is a personal account, outstanding rent is nominal account
2. In Journal Entries which pertain to outstanding entries, prepaid entries, depreciation entries are called  
a) Adjustment Entries      b) Rectification Entries  
c) Transfer Entries          d) closing Entries
3. In a three column cash book the discount columns are  
a) Totalled but not adjusted  
b) Totalled and adjusted  
c) Totalled but not balanced  
d) None of the above
4. Cash Book O.D. Balance Rs. 2,000. It was found that cheques of Rs. 100, Rs. 150, Rs. 175 which are issued but not presented till the date and the cheque of Rs. 600 deposited but have not been cleared, then O.D. Balance as per pass book is :  
a) Rs. 2150                      b) Rs. 2175  
c) Rs. 1475                      d) Rs. 1925
5. Bank Balance O.D. as per pass book Rs. 26,500. Calculate balance as per cash book is  
1) Cheques deposited for Rs. 4700 but not collected  
2) Cheques issued for Rs. 11,000 but not presented  
3) Bill discounted and dishonoured Rs. 4000, and Bank paid noting charges Rs. 200  
a) Rs.28600 O.D              b) Rs.16000 O.D  
c) Rs.24000 O.D              d) Rs.28500 O.D
6. Agreement of Trial balance is not a \_\_\_\_ proof of accuracy  
a) Submissive                  b) Inclusive  
c) Exhaustive                  d) Conclusive
7. Capital- Rs. 2,00,000      Interest paid- Rs. 2310  
Debtors- Rs. 15,200      Discount allowed- Rs. 820  
Creditors- Rs.12,960      Discount received- Rs. 1030  
Purchases- Rs. 92,670      Rent                      - Rs. 14,670  
Sales- Rs. 1,16,850      Loan                      - Rs. 12,060  
   Opening stock      - Rs. 56,000  
   Sales returns      - Rs. 27,430  
Debit Total of Trial Balance will be  
a) Rs. 2,09,000                  b) Rs. 2,09,100  
c) Rs. 2,10,000                  d) None
8. From the following information find the amount to be debited to P & L A/c for the period ending 31-03-2014. Provision for doubtful debts - Rs. 800 (on 01-04-2013)  
Debtors on 31-03-2014 - Rs. 40,000  
Bad debts                              - Rs. 2,000  
Bad debts to be written off and provision for doubtful debts is to be created @ 5% on debtors  
a) Rs. 3100                          b) Rs. 4000  
c) Rs. 3200                          d) Rs. 3900
9. Opening Stock Rs.30,000,  
Cost of goods available for sale Rs. 1,60,000,  
Sales were Rs. 1,60,000.  
Gross profit on sales is 30%  
Calculate closing stock.  
a) Nil                                  b) Rs. 48,000  
c) Rs. 98,000                      d) None of the above
10. By products are generally valued at \_\_\_\_ when the cost of by products is not directly traceable.  
a) Cost of main products  
b) N.R.V.  
c) Cost of main product or N.R.V. which ever is lower  
d) None of these
11. The purpose of Accommodation bill  
a) To facilitate trade transmission  
b) To finance the actual purchases & sales  
c) When both parties are in need of funds  
d) None of these

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**Liyong Dong**



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

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

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

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web aug 26 2023 derive chemical equations from narrative descriptions of chemical reactions write and balance chemical equations in molecular total ionic and net ionic formats an earlier chapter of this text introduced the use of element symbols to represent individual atoms

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web feb 19 2022 in a balanced chemical equation both the numbers of each type of atom and the total charge are the same on both sides equations ref 3 1 1 and ref 3 1 2 are balanced chemical equations what is different on each side of the equation is how the atoms are arranged to make molecules or ions

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web mar 14 2023 the ultimate goal for balancing chemical equations is to make both sides of the reaction the reactants and the products equal in the number of atoms per element this stems from the universal law of the conservation of mass which states that matter can neither be created nor destroyed

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stuff is reacting mass wise stoichiometry

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide  $\text{CaO}$  and  $\text{H}_2\text{O}$  to form aqueous calcium hydroxide  $\text{Ca(OH)}_2$  is it possible to balance the equation if you incorrectly identify the product as  $\text{CaOH}$  and if so what is the equation

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web balance a chemical equation when given the unbalanced equation explain the role of the law of conservation of mass in a chemical reaction even though chemical compounds are broken up and new compounds are formed during a chemical reaction atoms in the reactants do not disappear nor do new atoms appear to form the products

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web dec 2 2020 a balanced chemical equation tells you the amounts of reactants and products needed to satisfy the law of conservation of mass basically this means there are the same numbers of each type of atoms on the left side of the equation as there are on the right side of the equation

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web balance the following chemical equation  $\text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$  note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem do 4 problems

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web oct 13 2023 method 1 doing a traditional balance download article 1 write down your given equation for this example you will use  $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$  this reaction occurs when propane  $\text{C}_3\text{H}_8$  is burned in the presence of oxygen to produce water and carbon dioxide 2 write down the number of atoms per element

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