

Physical Chemistry

TOTAL MCQs= 100

1. If a chemical reaction is in equilibrium then which of the following is true?
 (a) reactants are completely transformed into products
 (b) formation of products is minimized
 (c) rate of forward and backward reactions are not equal
 (d) **none of these**
2. If K_1 and K_2 are rate constants for forward and backward reactions respectively then equilibrium constant is:
 (a) K_2/K_1
 (b) $K_1 \times K_2$
 (c) $K_1 + K_2$
 (d) **None of these**
3. Which of the following correctly represents the unit of K_p ?
 (a) $1/\text{atm}$
 (b) $(\text{atm})^2$
 (c) atm
 (d) **$(\text{atm})^{2\Delta n}$**
4. Relation between K_p and K_c is given by:
 (a) $K_p = K_c / (RT)^2$
 (b) $K_p = K_c / RT$
 (c) $K_p = K_c + RT^{\Delta n}$
 (d) **None of these**
5. At chemical equilibrium the value of change in Gibbs free energy ΔG is
 (a) greater than zero
 (b) **equal to zero**
 (c) smaller than zero
 (d) any value
6. Most effective catalysts are usually:
 (a) alkali metals
 (b) radioactive metals
 (c) alkaline earth metals
 (d) **None of these**
7. The name "catalyst" was proposed by which of the following scientist?
 (a) Rutherford
 (b) Langmuir
 (c) **Berzilius**
 (d) None of these
8. Catalyst Poison is essentially
 (a) Homogenous catalyst
 (b) Heterogenous catalyst
 (c) Autocatalyst
 (d) **An inhibitor**
9. in case of auto catalysis which of the following statements is true
 (a) reactant catalysis
 (b) solvent catalysis
 (c) heat produced in a reaction catalyzed the reaction
 (d) **none of these**
10. platinized asbestos is used in the manufacture of sulfuric acid. It is an example of
 (a) homogeneous catalysis
 (b) **heterogeneous catalysis**
 (c) induced catalysis
 (d) none of these
11. in the formation of ammonium by Habers process , Molybdenum is used as
 (a) negative catalyst
 (b) inhibitor
 (c) Homogenous catalyst
 (d) **None of these**
12. the catalyst used in Ziegler method for the polymerization of ethylene is:
 (a) $\text{ZnO} + \text{CuO}$
 (b) $\text{Cr}_2\text{O}_3 + \text{Fe}_2\text{O}_3$
 (c) Pt
 (d) **None of these**
13. In the formation of diethylether from ethanol the catalyst used is
 (a) NO
 (b) Ni
 (c) Fe/Mo
 (d) **Alumina**
14. catalyst used in the formation of methanol from water gas is:
 (a) $\text{CoO} + \text{ZnO} + \text{Cr}_2\text{O}_3$
 (b) $\text{CaO} + \text{ZnO} + \text{Cr}_2\text{O}_3$
 (c) **$\text{CuO} + \text{ZnO} + \text{Cr}_2\text{O}_3$**
 (d) None of these
15. which of the following is crystalline in nature?
 (a) glass
 (b) rubber
 (c) plastic
 (d) **none of these**
16. liquid crystals have no ordered structure of molecules.
 (a) smectic liquid crystals
 (b) **nematic liquid crystals**
 (c) cholesteric liquid crystals
 (d) none of these
17. Radius ratio for NaCl crystal is around:
 (a) **0.524**
 (b) 0.225
 (c) 0.781
 (d) 0.155

Physical Chemistry Mcqs

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