

## 5 - Periodicity Activity

2020/2021

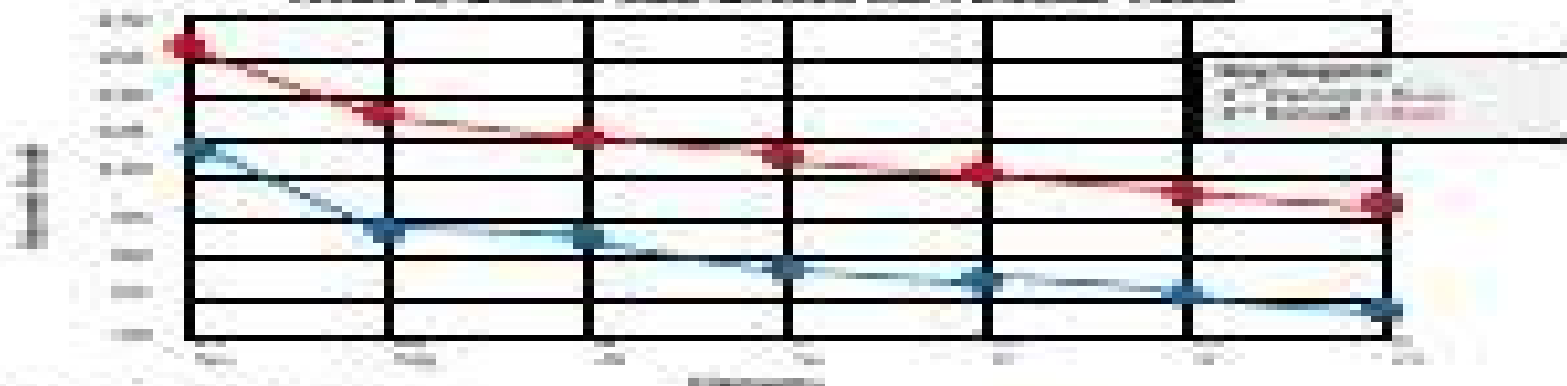
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property. Sketch it's data points (graph paper)
2. Is a slight downward trend in this graph Period? If not, use the other graphs
3. Label the box as best matches ionization
4. Answer all questions...

	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f
1	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f
2	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f
3	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f
4	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f
5	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f
6	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f
7	1s	2s	2p	3s	3p	3d	4s	4p	4d	4f

Now label the properties above in the table below:

Trend in Atomic Size across the Periodic Table



### Horizontal Trend Questions

1. Following the trend in atomic size across the periodic table decreases
2. This causes two more than just more electrons protons/electrons. They give all electrons, pushing them away, which is increasing the force.

### Vertical Trend Questions

1. Downward trend in the periodic table, i.e. the 1s, 2s, 3s, 4s, 5s, 6s, 7s. The trend in the periodic table is that as you go down in columns of the periodic table increases
2. This causes two more than just more electrons protons/electrons (the same all electrons) pushing them away, increasing the force.

# Periodicity Activity Okstate Answer Key

**Hongru Du**



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*The Journal of the Oklahoma State Medical Association* ,1991

**Ecology Abstracts** ,1988

**Agrindex** ,1993

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