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**Exceed to Succeed**

## **Worksheet: Periodic Trends**

### **1. ATOMIC RADIUS**

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li, Na, K
- c. Ge, P, O
- d. C, N, Al
- e. Al, Cl, Ga

### **2. IONIC RADIUS**

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a.  $\text{Mg}^{2+}$ ,  $\text{Si}^{4+}$ ,  $\text{S}^{2-}$
- b.  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Ba}^{2+}$
- c.  $\text{F}^-$ ,  $\text{Cl}^-$ ,  $\text{Br}^-$
- d.  $\text{Ba}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Zn}^{2+}$
- e.  $\text{Si}^{4+}$ ,  $\text{P}^{3-}$ ,  $\text{O}^{2-}$

### **3. IONIZATION ENERGY**

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F, Cl, Br
- d. Ba, Cu, Ne
- e. Si, P, He

### **4. ELECTRONEGATIVITY**

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C, O, Ne
- c. Si, P, O
- d. K, Mg, P
- e. S, F, He

# Periodic Trends Worksheet 1 Using The Periodic Law Answers

**A Gutmann**



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## **Table of Contents Periodic Trends Worksheet 1 Using The Periodic Law Answers**

1. Understanding the eBook Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - The Rise of Digital Reading Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Personalized Recommendations
  - Periodic Trends Worksheet 1 Using The Periodic Law Answers User Reviews and Ratings

- Periodic Trends Worksheet 1 Using The Periodic Law Answers and Bestseller Lists
- 5. Accessing Periodic Trends Worksheet 1 Using The Periodic Law Answers Free and Paid eBooks
  - Periodic Trends Worksheet 1 Using The Periodic Law Answers Public Domain eBooks
  - Periodic Trends Worksheet 1 Using The Periodic Law Answers eBook Subscription Services
  - Periodic Trends Worksheet 1 Using The Periodic Law Answers Budget-Friendly Options
- 6. Navigating Periodic Trends Worksheet 1 Using The Periodic Law Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Trends Worksheet 1 Using The Periodic Law Answers Compatibility with Devices
  - Periodic Trends Worksheet 1 Using The Periodic Law Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Highlighting and Note-Taking Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Interactive Elements Periodic Trends Worksheet 1 Using The Periodic Law Answers
- 8. Staying Engaged with Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Trends Worksheet 1 Using The Periodic Law Answers
- 9. Balancing eBooks and Physical Books Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Trends Worksheet 1 Using The Periodic Law Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Setting Reading Goals Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Fact-Checking eBook Content of Periodic Trends Worksheet 1 Using The Periodic Law Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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