

Periodic Table: Virtual Lab Document



Part 1:

Follow the instructions in the Periodic Table Virtual Lab Lesson to color the periodic table.

Click on the image below -> edit -> drop down arrow beside the line -> select tool -> change colors.

H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Uun	Uuu	Uub						
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

Part 2: Go to chemicalelements.com.

1. Click on Alkali Metals (left bar) to answer the following questions:

- What is the group number? [1]
- Are these metals reactive? [Yes, they are highly reactive]
- Do these metals occur freely in nature? [No, they don't occur freely in nature, but they are found in various minerals.]
- How many electrons are in their outer shell? [1]
- What are the three characteristics of ALL metals? [Shiny luster, good conductors of heat and electricity, malleable and ductile.]
- Are these metals soft or hard? [soft]
- Name the two most reactive elements in this group? [Potassium and Sodium]
- What happens when they are exposed to water? [They react vigorously with water, producing hydrogen gas and an alkaline solution. The reaction can be violent and can even ignite the hydrogen gas produced.]

Periodic Table Grouping Lab Answer Key

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Periodic Table Grouping Lab Answer Key

1. Understanding the eBook Periodic Table Grouping Lab Answer Key
 - The Rise of Digital Reading Periodic Table Grouping Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Grouping Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Periodic Table Grouping Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Grouping Lab Answer Key
 - Personalized Recommendations
 - Periodic Table Grouping Lab Answer Key User Reviews and Ratings

- Periodic Table Grouping Lab Answer Key and Bestseller Lists
- 5. Accessing Periodic Table Grouping Lab Answer Key Free and Paid eBooks
 - Periodic Table Grouping Lab Answer Key Public Domain eBooks
 - Periodic Table Grouping Lab Answer Key eBook Subscription Services
 - Periodic Table Grouping Lab Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table Grouping Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Grouping Lab Answer Key Compatibility with Devices
 - Periodic Table Grouping Lab Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Grouping Lab Answer Key
 - Highlighting and Note-Taking Periodic Table Grouping Lab Answer Key
 - Interactive Elements Periodic Table Grouping Lab Answer Key
- 8. Staying Engaged with Periodic Table Grouping Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Grouping Lab Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Grouping Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Grouping Lab Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Grouping Lab Answer Key
 - Setting Reading Goals Periodic Table Grouping Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Grouping Lab Answer Key
 - Fact-Checking eBook Content of Periodic Table Grouping Lab Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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