QUIZIZZ NAME: Chemical Bonds & Intermolecular Forces CLASS: 22 Questions DATE:

1. Identify the phrases that generally apply to molecular compounds,

A c. Use covalent bonds a. Are made of non-metals chemically joined together.

d. Use ionic bonds De. Have high boiling points

E b. Have low melting points

The attraction of the free-floating valence electrons for the positively charged metal ions.

A lonic bond B Metallic Bond

C London dispersion forces D Covalent Bond

Which of the following elements will form a similar ion?

A Be, Ca, Ba B Na, K, Li

C Cl, Ar, Br D F, Cl, Br

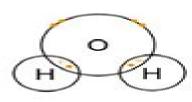
4. Which of the following is true about ionic bonding

A It is weaker than the covalent bond. B Electrons are donated from one element to another

Is made by the attraction between a cation and an anion.

It is mostly formed between elements from groups IA, IIA, IIIA and a nonmetal.

What type of bond is formed between the hydrogen and oxygen atoms in this image?



A Metallic bond

5

B London Dispersion Forces

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