

QUESTION 4 (Start on a new page)

The reaction between zinc and dilute hydrochloric acid is represented by the equation below.



During an experiment, 1,5 g of zinc reacts with excess dilute hydrochloric acid to produce hydrogen gas at STP.

- 4.1 What is represented by the acronym *STP*?
- 4.2 Define the term *one mole of a substance*.
- 4.3 Calculate the:
 - 4.3.1 Number of moles of zinc used
 - 4.3.2 Volume (in cm^3) of hydrogen gas produced at STP
 - 4.3.3 Number of chlorine atoms present in the ZnCl_2 produced

QUESTION 5 (Start on a new page)

- 5.1 What is the difference between a vector quantity and a scalar quantity?
- 5.2 List any TWO scalar quantities AND TWO vector quantities that you know.
- 5.3 **Learner A** applies a force of 20 N on **Learner B**. The direction of the force is due west. At the same time **Learner C** applies a force of 30 N on **Learner B** in the OPPOSITE direction.

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Albert A Gayle



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