

PERIODIC TABLE

Atomic Properties of the Elements

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					2
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd					
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																									
speed of light in vacuum c = 299 792 458 m s ⁻¹ (exact)																									
Planck constant h = 6.626 070 15 × 10 ⁻³⁴ J s (exact)																									
elementary charge e = 1.602 176 634 × 10 ⁻¹⁹ C																									
electron mass m _e = 9.109 383 56 × 10 ⁻³¹ kg																									
proton mass m _p = 1.672 621 9 × 10 ⁻²⁷ kg																									
fine-structure constant α = 1/137.036																									
Rydberg constant R _∞ = 10 973 731.568 162 m ⁻¹																									
Bohr magneton μ _B = 9.274 010 078 3 × 10 ⁻²⁴ J K ⁻¹																									
Boltzmann constant k = 1.380 658 × 10 ⁻²³ J K ⁻¹																									

Periodic Table Chemistry If8766

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