

# Periodic Law

| Periodic Table of the Elements   |          |  |            |            |            |            |            |            |            |           |           |  |           |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
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| 1<br>H   |          |  |            |            |            |            |            |            |            |           |           |  |           |          |          |          | 2<br>He  |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
| 3<br>Li  | 4<br>Be  | <div><div>hydrogen</div><div>alkali metals</div><div>alkali earth metals</div><div>transition metals</div></div> |            |            |            |            |            |            |            |           |           | <div><div>poor metals</div><div>metalloids</div><div>noble gases</div><div>rare earth metals</div></div> |           |          |          |          |          | 5<br>B   | 6<br>C   | 7<br>N   | 8<br>O   | 9<br>F   | 10<br>Ne |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
| 11<br>Na   | 12<br>Mg |  |            |            |            |            |            |            |            |           |           | 13<br>Al   | 14<br>Si  | 15<br>P  | 16<br>S  | 17<br>Cl | 18<br>Ar |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
| 19<br>K  | 20<br>Ca | 21<br>Sc   | 22<br>Ti   | 23<br>V    | 24<br>Cr   | 25<br>Mn   | 26<br>Fe   | 27<br>Co   | 28<br>Ni   | 29<br>Cu  | 30<br>Zn  | 31<br>Ga   | 32<br>Ge  | 33<br>As | 34<br>Se | 35<br>Br | 36<br>Kr |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
| 37<br>Rb   | 38<br>Sr | 39<br>Y  | 40<br>Zr   | 41<br>Nb   | 42<br>Mo   | 43<br>Tc   | 44<br>Ru   | 45<br>Rh   | 46<br>Pd   | 47<br>Ag  | 48<br>Cd  | 49<br>In   | 50<br>Sn  | 51<br>Sb | 52<br>Te | 53<br>I  | 54<br>Xe |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
| 55<br>Cs   | 56<br>Ba | 57<br>La   | 72<br>Hf   | 73<br>Ta   | 74<br>W    | 75<br>Re   | 76<br>Os   | 77<br>Ir   | 78<br>Pt   | 79<br>Au  | 80<br>Hg  | 81<br>Tl   | 82<br>Pb  | 83<br>Bi | 84<br>Po | 85<br>At | 86<br>Rn |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
| 87<br>Fr   | 88<br>Ra | 89<br>Ac   | 104<br>Unq | 105<br>Unp | 106<br>Unh | 107<br>Uns | 108<br>Uno | 109<br>Une | 110<br>Uun |           |           |  |           |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |
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| 90<br>Th   | 91<br>Pa | 92<br>U  | 93<br>Np   | 94<br>Pu   | 95<br>Am   | 96<br>Cm   | 97<br>Bk   | 98<br>Cf   | 99<br>Es   | 100<br>Fm | 101<br>Md | 102<br>No  | 103<br>Lr |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |          |         |          |          |          |          |          |          |          |           |           |           |           |

# Periodic Law Lab 17 Answer Key

**Yicheng Fang**



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