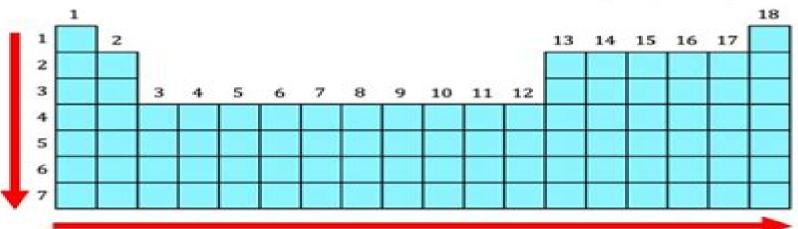
Periodic Trends - Atomic Mass

- The atomic mass describes the mass of each individual element.
- Atomic mass increases moving down each group and right across a period.

<u>Instructions</u>: In the periodic table below, draw two arrows representing the direction that atomic mass <u>increases</u> across groups and periods.



Instructions: In each question, identify the element with the greatest atomic mass:

1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F)

lodine

2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B)

Oxygen

3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe)

Xenon

Instructions: In each question, list the elements from least to greatest atomic mass:

4) Magnesium (Mg), Beryllium (Be), Barium (Ba)

Be, Mg, Ba

5) Gallium (Ga), Selenium (Se), Potassium (K)

K, Ga, Se

Periodic Trends In Atomic Properties Answers

Robert R. Redfield

Periodic Trends In Atomic Properties Answers:

Enjoying the Tune of Phrase: An Mental Symphony within **Periodic Trends In Atomic Properties Answers**

In a world eaten by screens and the ceaseless chatter of quick interaction, the melodic beauty and mental symphony produced by the prepared term frequently diminish in to the back ground, eclipsed by the persistent sound and disturbances that permeate our lives. But, located within the pages of **Periodic Trends In Atomic Properties Answers** a marvelous fictional treasure overflowing with raw feelings, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that interesting masterpiece conducts visitors on a mental trip, well unraveling the hidden melodies and profound affect resonating within each cautiously constructed phrase. Within the depths of this poignant examination, we shall discover the book is central harmonies, analyze their enthralling writing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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