

Name: _____ () Class: _____ Date: _____

The Periodic Table**General Properties of the Periodic Table**

1. The Periodic Table is a table that groups elements according to certain physical properties or chemical properties.
2. The elements are arranged in order of **increasing proton number (atomic number)**.
3. A **HORIZONTAL** row of elements is known as a **Period**.
Across a period, the element changes from **metal** character to **non-metal** character.
Elements on left hand side of Periodic Table are _____.
Elements on right hand side of Periodic Table are _____.
4. A **VERTICAL** row of elements is known as a **Group**.
5. Hydrogen and Helium are in the same **Period 1**.
They all have _____ **electron shell**.

Draw out the electronic structure of Hydrogen and Helium below:



Hydrogen



Helium

6. The following elements are in the same **Period 2**.
They all have _____

Draw out the electronic structure of these elements.

--	--	--	--	--	--	--	--

Lithium

Beryllium

Boron

Carbon

Nitrogen

Oxygen

Fluorine

Neon

7. Lithium is in **Group 1**. Lithium has _____ outer electron.
Beryllium is in **Group 2**. Beryllium has _____ outer electrons.
Oxygen is in **Group 6**. Oxygen has _____

Quick Check

Without reference to the Periodic Table, predict the Period and Group of the elements below:

- | | | | |
|-------------------------|-------|-----------------------------|-------|
| a) Ca – 2.8.8.2 | _____ | d) Xe 2.8.18.18.8 | _____ |
| b) S – 2.8.6 | _____ | e) Pb 2.8.18.32.18.4 | _____ |
| c) Ga – 2.8.18.3 | _____ | f) Rb – 2.8.18.8.1 | _____ |

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Spotlight Science Keith Johnson, Sue Adamson, Gareth Williams, 2002 This Spiral Edition Teacher Support Pack offers comprehensive support and guidance providing the best possible learning experience for your students and saving time for everyone in the department

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