

Name: _____

Dates:

CHEMISTRY

Periodic Table of the Elements

Use the Periodic Table of the Elements to answer the questions.

The image shows a standard periodic table of elements. The elements are color-coded by groups:

- Group 1 (Alkali metals): Light blue
- Group 2 (Alkaline earth metals): Orange
- Groups 3-10 (Transition metals): Yellow
- Group 11 (Coinage metals): Light green
- Group 12 (Zn, Cd, Hg): Dark green
- Group 13 (Boron group): Light blue
- Group 14 (Carbon group): Blue
- Group 15 (Nitrogen group): Purple
- Group 16 (Chalcogens): Pink
- Group 17 (Halogens): Red
- Group 18 (Noble gases): Light yellow
- Period 7 elements (Fr, Ra, Ac-Lr): Dark blue
- Lanthanide and Actinide series (bottom): Dark blue

Labels for Boron (B) and its isotopes are present:

- B**: Element symbol for Boron, located in Group 13, Period 2.
- Boron**: Element name, located below the symbol B.
- Atomic number**: 5, located to the left of the symbol B.
- Atomic mass**: 10.811, located to the right of the symbol B.
- Isotopes**: A label pointing to the isotopes section, located to the right of the symbol B.
- Group**: A label pointing to the group number, located to the right of the symbol B.
- Period**: A label pointing to the period number, located to the right of the symbol B.

Isotopes of Boron are listed in the top right corner:

- ^{10}B** : Stable isotope, 19.9% abundance.
- ^{11}B** : Stable isotope, 80.1% abundance.
- ^{12}B** : Radioactive isotope, 0.0% abundance.
- ^{13}B** : Radioactive isotope, 0.0% abundance.
- ^{14}B** : Radioactive isotope, 0.0% abundance.
- ^{15}B** : Radioactive isotope, 0.0% abundance.
- ^{16}B** : Radioactive isotope, 0.0% abundance.
- ^{17}B** : Radioactive isotope, 0.0% abundance.
- ^{18}B** : Radioactive isotope, 0.0% abundance.
- ^{19}B** : Radioactive isotope, 0.0% abundance.
- ^{20}B** : Radioactive isotope, 0.0% abundance.
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- ^{67}B** : Radioactive isotope, 0.0% abundance.
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- ^{77}B** : Radioactive isotope, 0.0% abundance.
- ^{78}B** : Radioactive isotope, 0.0% abundance.
- ^{79}B** : Radioactive isotope, 0.0% abundance.
- ^{80}B** : Radioactive isotope, 0.0% abundance.
- ^{81}B** : Radioactive isotope, 0.0% abundance.
- ^{82}B** : Radioactive isotope, 0.0% abundance.
- ^{83}B** : Radioactive isotope, 0.0% abundance.
- ^{84}B** : Radioactive isotope, 0.0% abundance.
- ^{85}B** : Radioactive isotope, 0.0% abundance.
- ^{86}B** : Radioactive isotope, 0.0% abundance.
- ^{87}B** : Radioactive isotope, 0.0% abundance.
- ^{88}B** : Radioactive isotope, 0.0% abundance.
- ^{89}B** : Radioactive isotope, 0.0% abundance.
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- ^{91}B** : Radioactive isotope, 0.0% abundance.
- ^{92}B** : Radioactive isotope, 0.0% abundance.
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- ^{94}B** : Radioactive isotope, 0.0% abundance.
- ^{95}B** : Radioactive isotope, 0.0% abundance.
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- ^{97}B** : Radioactive isotope, 0.0% abundance.
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- ^{100}B** : Radioactive isotope, 0.0% abundance.
- ^{101}B** : Radioactive isotope, 0.0% abundance.
- ^{102}B** : Radioactive isotope, 0.0% abundance.
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- ^{105}B** : Radioactive isotope, 0.0% abundance.
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- ^{113}B** : Radioactive isotope, 0.0% abundance.
- ^{114}B** : Radioactive isotope, 0.0% abundance.
- ^{115}B** : Radioactive isotope, 0.0% abundance.
- ^{116}B** : Radioactive isotope, 0.0% abundance.
- ^{117}B** : Radioactive isotope

Part 1: Atomic Number. On the line next to the element, write the atomic number of the element. Your answer should be 1, 2, or 3 digits. No letters.

- | | | | | | |
|----|--|-----------|-----|--|--------------|
| 1. | | Hydrogen | 7. | | Cerium |
| 2. | | Oxygen | 8. | | Iridium |
| 3. | | Magnesium | 9. | | Bismuth |
| 4. | | Vanadium | 10. | | Uranium |
| 5. | | Krypton | 11. | | Fermium |
| 6. | | Ruthenium | 12. | | Darmstadtium |

Periodic Table Answers Section 3

J Dewey



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Table of Contents Periodic Table Answers Section 3

1. Understanding the eBook Periodic Table Answers Section 3
 - The Rise of Digital Reading Periodic Table Answers Section 3
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Answers Section 3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Answers Section 3
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Answers Section 3
 - Personalized Recommendations
 - Periodic Table Answers Section 3 User Reviews and Ratings
 - Periodic Table Answers Section 3 and Bestseller Lists
5. Accessing Periodic Table Answers Section 3 Free and Paid eBooks
 - Periodic Table Answers Section 3 Public Domain eBooks
 - Periodic Table Answers Section 3 eBook Subscription Services
 - Periodic Table Answers Section 3 Budget-Friendly Options

6. Navigating Periodic Table Answers Section 3 eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Answers Section 3 Compatibility with Devices
 - Periodic Table Answers Section 3 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Answers Section 3
 - Highlighting and Note-Taking Periodic Table Answers Section 3
 - Interactive Elements Periodic Table Answers Section 3
8. Staying Engaged with Periodic Table Answers Section 3
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Answers Section 3
9. Balancing eBooks and Physical Books Periodic Table Answers Section 3
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Answers Section 3
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Answers Section 3
 - Setting Reading Goals Periodic Table Answers Section 3
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Answers Section 3
 - Fact-Checking eBook Content of Periodic Table Answers Section 3
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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