

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-3)

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction.

Excess: All reagents except the limiting reagent will be present in excess. Excess is the amount of reagent that is not used in the reaction.

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction.

Excess: All reagents except the limiting reagent will be present in excess.



The reaction is balanced. The limiting reagent is the reactant that will completely run out first in a chemical reaction.

2H ₂	20.0 g
O ₂	16.0 g
H ₂ O	36.0 g
H ₂	10.0 g

- 20.0 g of H₂ and 16.0 g of O₂ are used. The limiting reagent is H₂. The excess reagent is O₂. The limiting reagent is the reactant that will completely run out first in a chemical reaction.



- 20.0 g of H₂ and 16.0 g of O₂ are used. The limiting reagent is H₂. The excess reagent is O₂. The limiting reagent is the reactant that will completely run out first in a chemical reaction.

- 20.0 g of H₂ and 16.0 g of O₂ are used. The limiting reagent is H₂. The excess reagent is O₂. The limiting reagent is the reactant that will completely run out first in a chemical reaction.



- 20.0 g of H₂ and 16.0 g of O₂ are used. The limiting reagent is H₂. The excess reagent is O₂. The limiting reagent is the reactant that will completely run out first in a chemical reaction.



Percent Yield And Limiting Reactants Quiz

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