

Group name*		
Hydrogen & the alkali metals	14	Crystallogens
Alkaline earth metals	15	Psithogens
Coinage metals	16	Chalcogens
Volatile metals	17	Halogens
Isotopans	18	Helium atom

*Groups 1 to 20 are ranked after their first matches. i.e. Group 1 is the top-ranked group.

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☐ **Radioactive**
 Element is
 likely to be either
 a metalloid or a
 good transition
 metal

1000

Moderate to high density metals, with high melting and boiling points, many have high hardness and mechanical strength. Those from R1 onwards are synthetic. Chemically show variable valency and strong tendency to form coordination and colored compounds. Oxides are basic, amphoteric or acidic, depending on oxidation state.

100

Resistant to corrosion and oxidation, unlike base metals such as Fe, Ni, Cu and Zn. Commonly found together in native form. Good catalytic properties. Very easily electronegative, oxidises easily.

Keywords: *work engagement, organizational commitment, turnover intentions, organizational citizenship behaviors, organizational identification*

Soft (or brittle), low strength, and with melting points lower than transition metals. Crystalline structures tend to show directional bonding, with generally greater complexity or fewer nearest neighbours than other metals. All used or treated as soft metal lubricants. Most are chemically weak; a minority are moderately electropositive. Characterised, to varying degrees, by covalent bonding tendencies, anaphoreticity, and anionic species such as triphenyl, siloxides, and boronfluorides.

N_2 , O_2 , and O_3 are, respectively, colorless, pale yellow, and pale yellow-green gases; Br_2 is a red-brown liquid; I_2 is a metallic looking solid. Chemically iodine has to have high ionisation energies, electron affinities, and electronegativity values, and be relatively strong oxidising agents. Manifestations include oxygen's major association with the ubiquitous processes of corrosion and combustion, and the intrinsically corrosive nature of the non-metallic halogens. All four tend to form predominantly ionic compounds with metals. Their oxides are acidic.

100

Complex, colorless, non-flammable gases with very low chemical reactivity. The first noble gas compound, of approximate composition XePtF_6 , was not prepared until 1962. More than half-a-thousand noble gas compounds were synthesized in the ensuing half-century. Compounds of He, Ne and Ar, in ambient conditions, are not known.

[illegible]

- 1/4 strength, average fluid of groups 1-3
- 1/4 strength, more average fluid of groups 4-10

Periodic Table Unscrambled Table Answers

M Lipman



Periodic Table Unscrambled Table Answers:

Periodic Table Unscrambled Table Answers Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the power of words has be evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Periodic Table Unscrambled Table Answers**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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