Active mobile

Pre-transition metals

Light, reactive, mostly of the mechanical strength, mailing and boling points. The group 1 members are easily out with a traffic. Only Se, Mg and Alfhane any structural years. Most form basic points (Se and W are amythology).

And section

Closely recentive Ca. for and Ba, but many are hower, and they furn mostly pale coloured compounds. Similar to one another, e.g., Ca to Cu; jobs the fundamised, and hard to expansite for that reason. Abundant but writing dispersed, commercially waters concentrations are have. Contact are strongly to moderately haste. Sr. Ys. and Cu; waterfor west anotherisms.

Artinian

Soft, dense and reactive metals, those from Els onwers's are symbotic. Early activides (Th-Am) show some similarities to transition metals and have beaut, amphitisms, or acido cascies; use activides are more like terrhandes. Many activides from coloured compounds.

Proposition metals

Moderate to high density metals, with high melting and boiling points, many have high hardware and mechanical strength. Those from RF onwards are synthetic. Chamically show vertible valency and strong lendercy to form coordination and colored compounds. Oxides are basic, amphatenc or action, depending on calcilation state.

CONTRACTOR AND ADDRESS.

Parentant to common and oxidation, unlike base metals such as Fe, M., Cu and Jh. Commonly Yound together in native form. Good outsiylic properties. Titry wealthy sectropositive; small bits cooles.

The Properties and the

Suit or britise), low strength, and with meting points lower than transition messes. Crystalline structures hand to show directional bonding, with generally greater complexity or lewer neutral heightours than other metias. All used or trained as soft metal lubricants. Most are chemically weak, a minority are moderately electrogositive. Characterised, to varying degrees, by consideral bonding landernies, amphiciation, and extend species such as direction, strengths, and bemultures.

Microla I Novembrain

Siletalic looking brittle, fair electrical conductors. All known in semiconducting forms. Motivate lonisation energies (750-1,000 kJ/mol), and electroregativity values (1.9-2.2). Mostly behave chemically as non-metals. Can form alloys with metals. Amphibiant or wealth acidic oxides.

Internacials suppossible

If and N are optioned games, C. P*, and Se are metallic looking solds while S is yettin. The latter four are brittle but all known in malestee, plastic or ductile forms. Moderate to high consistion energies, low to high electron affection, moderate to high electronegativity values, retainely poor to moderately strung conting agents, tend to form obsessed or metallic to near-metallic compounds with metals. Their condex are acids, or needed third, C.O. NO, No.70.

" as block if the most codes form in problems conditions.

Green II

The least as 50 ft., and that biturates the a Larbound-Le. Carlot, and a to beautiful a. Carlot. Transit gamp their Sc.Y. and the

a. Labourt, represe hose of grove 1-2

Ly franch, more resenting those of groups it (II).

Compales nonmetals

Q. F. and G-are, respectively, coloriers, pair yellow, and pair jeffow-green passes. Br is ned-brown liquid I is a metallic looking solid. Chemically land to have to high lookinshon energies, election affinities, and electronegalishity values, and be relatively strong padding agents. Manifestations include oxygen's major association with the uniquipus processes of committee and combustion, and the introducially committee nature of the non-metallic hatogens. All the tend is form predominately look and processes of the non-metallic hatogens. All the tend is form predominately looks.

1 Nobile process

Colorless, odorless, non-flammable gases with very low chemically reactivity. The first notice gas compound, of approximate composition NaPPL, was not prepared until 1962. None than half-e-floward notice gas compounds were synthesised in the ensuing half-century. Compounds of the Ne, and Ar. in ambient conditions, see not known.

Periodic Table Guide Answer Sheet

Leo J. Malone, Theodore O. Dolter

Periodic Table Guide Answer Sheet:

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