# Percent, Actual, and Theoretical Yield

LiOH + KCI → LiCI + KOH

a) I began this reaction with 20 grams of lithium hydroxide. What is my theoretical yield of lithium chloride?

- b) I actually produced 6 grams of lithium chloride. What is my percent yield?
- 2) C<sub>3</sub>H<sub>6</sub> + 5 O<sub>2</sub> → 3 CO<sub>2</sub> + 4 H<sub>2</sub>O

a) If I start with 5 grams of C<sub>2</sub>H<sub>6</sub>, what is my theoretical yield of water?

- b) I got a percent yield of 75% How many grams of water did I make?
- Be + 2 HCl → BeCl<sub>2</sub> + H<sub>2</sub>

My theoretical yield of beryllium chloride was 10.7 grams. If my actual yield was 4.5 grams, what was my percent yield?

4) 2 NaCl + CaO → CaCl<sub>2</sub> + Na<sub>2</sub>O

What is my theoretical yield of sodium oxide if I start with 20 grams of calcium oxide?

# **Percent Actual And Theoretical Yield Answer Key**

Ralph H. Petrucci, William S. Harwood

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