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 $3\text{Cl}_2 + 3\text{KClO}_3 \rightarrow 3\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{OH}^- + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2 + 2\text{Al} + 3\text{K}_2\text{SO}_4 + 6\text{KBr} + 1\text{Al}_2\text{SO}_4 + 3\text{H}_2\text{O} + 2\text{CO}_2$

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 $\text{N}_2 + \text{F}_2 \rightarrow 2\text{NF}_3$
 $2\text{C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SNO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{H}_2\text{O} + 2\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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 $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $3\text{HBr} + 1\text{KHCO}_3 \rightarrow 1\text{H}_2\text{O} + 1\text{KBr} + \text{CO}_2$
 $2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow 1\text{Ga}_2\text{SO}_3 + 3\text{H}_2\text{O} + 2\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$ using the following equation $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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assuming ce cucl_2 cucx_2 is available in excess

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 $\text{br}_2 + \text{c}_6\text{h}_{10} + \text{o}_2 \rightarrow \text{co}_2 + \text{h}_2\text{o}$
 $\text{br}_3 + \text{hbr} + \text{khco}_3 \rightarrow \text{h}_2\text{o} + \text{kbr} + \text{co}_2$
 $\text{br}_4 + \text{gabr}_3 + \text{na}_2\text{so}_3 \rightarrow \text{ga}_2\text{so}_3 + \text{nabr}$
 $\text{br}_5 + \text{sno} + \text{nf}_3 \rightarrow \text{snf}_2 + \text{n}_2\text{o}_3$
 using the following equation
 $\text{br}_2 + \text{naoh} + \text{h}_2\text{so}_4 \rightarrow \text{h}_2\text{o} + \text{na}_2\text{so}_4$

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 $\text{hbr} + \text{khco}_3 \rightarrow \text{h}_2\text{o} + \text{kbr} + \text{co}_2$
 $\text{gabr} + \text{na}_2\text{so}_3 \rightarrow \text{ga}_2\text{so}_3 + \text{nabr}$
 $\text{sno} + \text{nf}_3 \rightarrow \text{snf}_2$

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 $\text{ce sio}_2 + 3\text{c} \rightarrow \text{sic} + 2\text{co}$

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 $\text{hbr} + \text{khco}_3 \rightarrow \text{h}_2\text{o} + \text{kbr} + \text{co}_2$
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